

HIGH TEMPERATURE OXIDATION OF E110 ALLOY WITH FeCrNi AND CrNi BASED COATINGS

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Introduction



First stage of all ATF-programs including ROSATOM – is **modification and coatings deposition** on Zr (E110) claddings.



ROSATOM

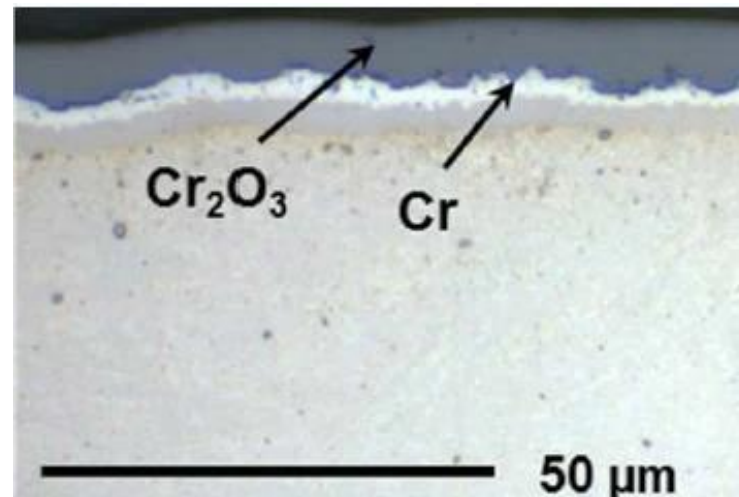
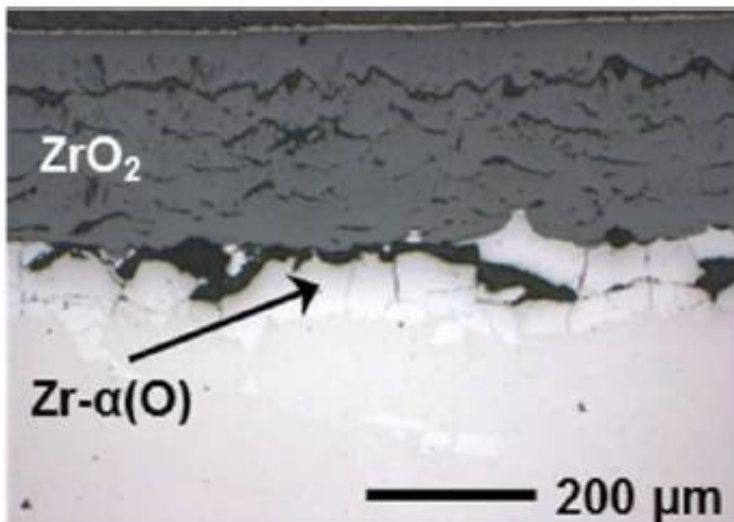


Korea Atomic Energy
Research Institute

Introduction



Cr-based coatings are investigated actively because of dense Cr_2O_3 layer forming during high-temperature oxidation.



Zry-4 uncoated (left) and Cr-coated (right) after steam oxidation 1200 °C 15000 s (P. Guillermier – IAEA 2015)

The main goal



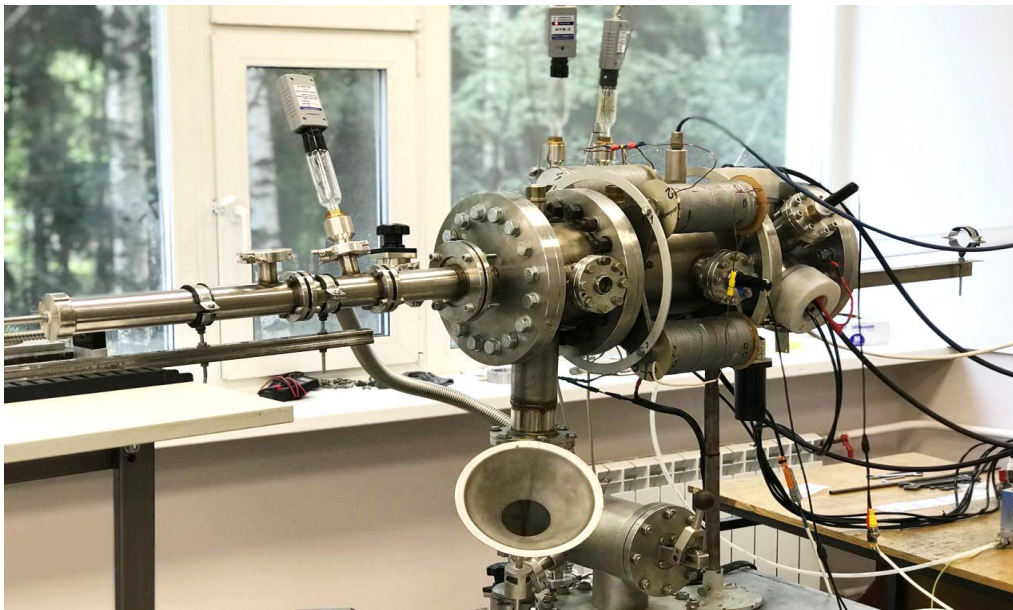
To investigate the effectiveness of **anticorrosion Cr-content coatings** on fuel claddings from E110 alloy

1. Batch of samples were prepared:
 \varnothing 9.1 / 9.5 mm, 10-500 mm length
 total amount about 300
2. Corrosion tests were performed:
 autoclave
 LOCA conditions
3. Material state was analyzed
4. **Fuel assemblies with coated samples were made for tests in MIR research reactor (now in progress)**

Coatings deposition



Coatings were deposited on claddings outer surface by ion-plasma methods.

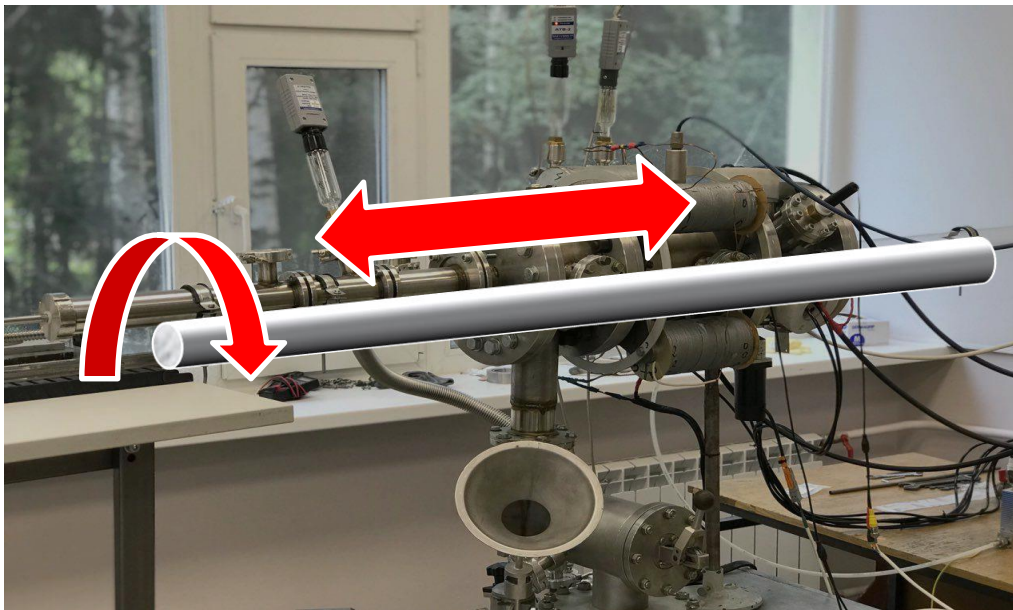


Installations for claddings ion-plasma treatment ILUR-03 (up) and KVK-10 (right)

Coatings deposition



Coatings homogeneity provided by technology: translational motion and axial rotation of samples.



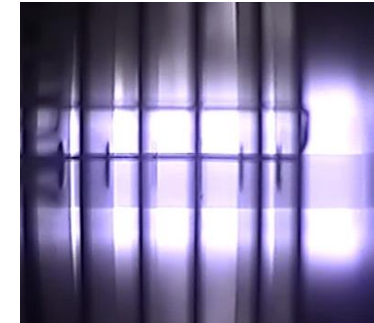
Installations for claddings ion-plasma treatment ILUR-03 (up) and KVK-10 (right)

Coatings deposition

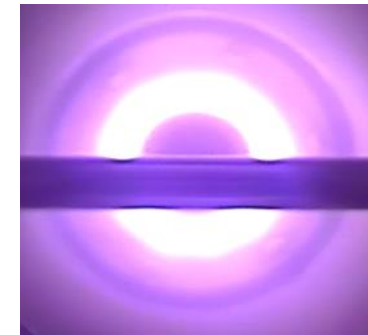


Coatings were deposited by several stages:

1. Pre-treatment by Ar^+ $E = 0.1\text{-}1.0$ keV



2. Multilayer magnetron deposition

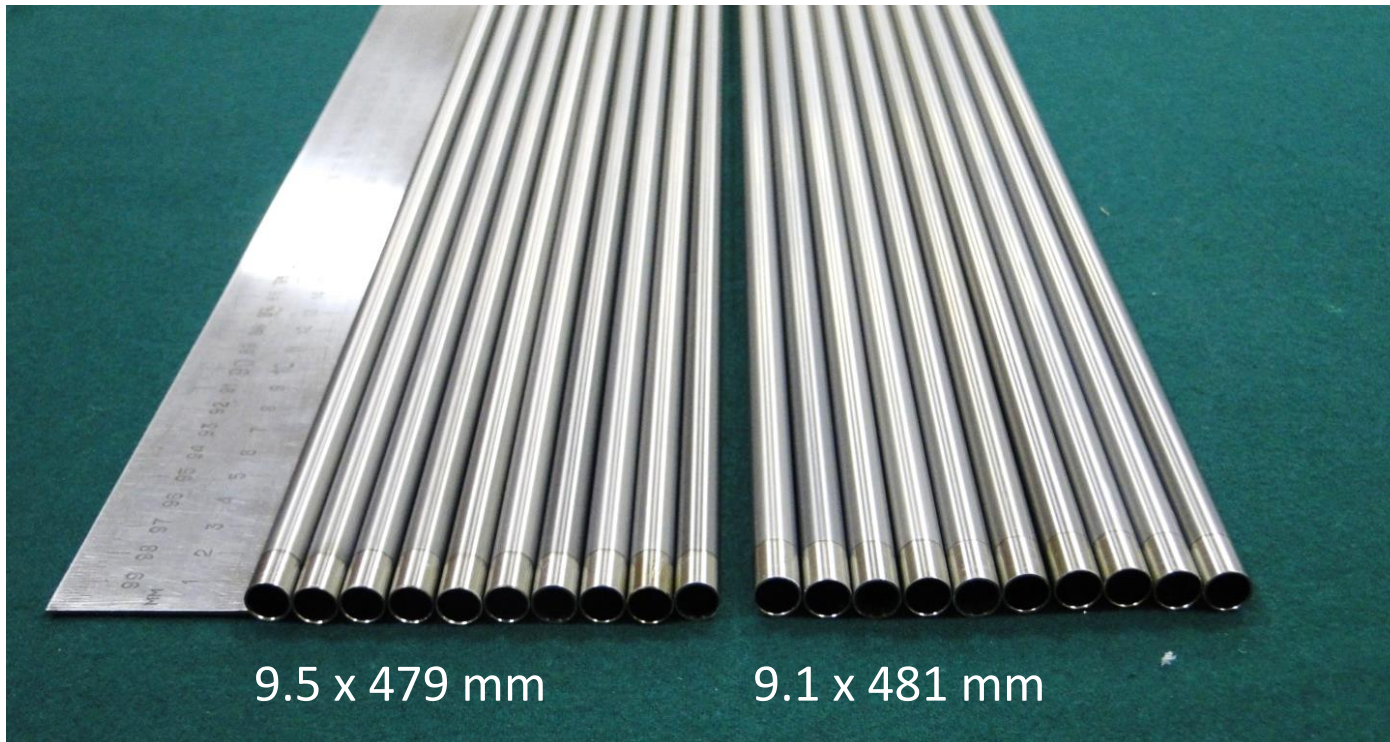


* Additional bombardment by Ar^+ $E = 35$ keV
(on KVK-10 only)

Coatings deposition



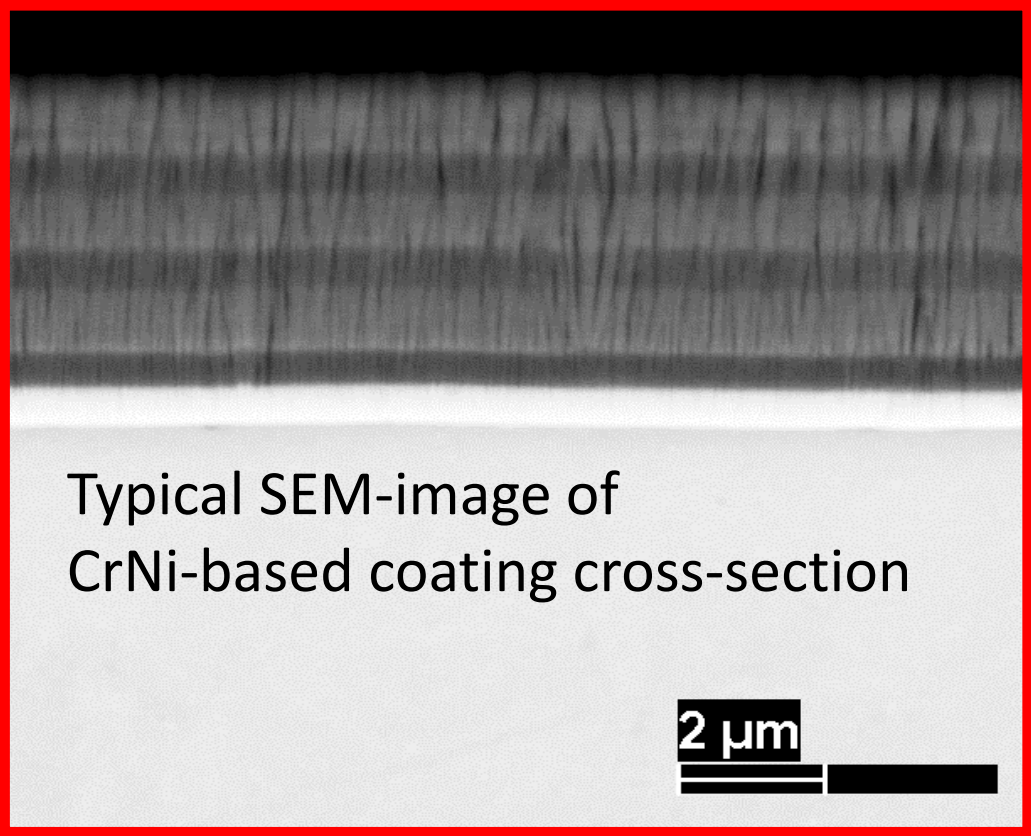
Coating base	Max thickness, μm
Cr	10 ± 1
FeCrNi	7 ± 1
NiCr	10 ± 1



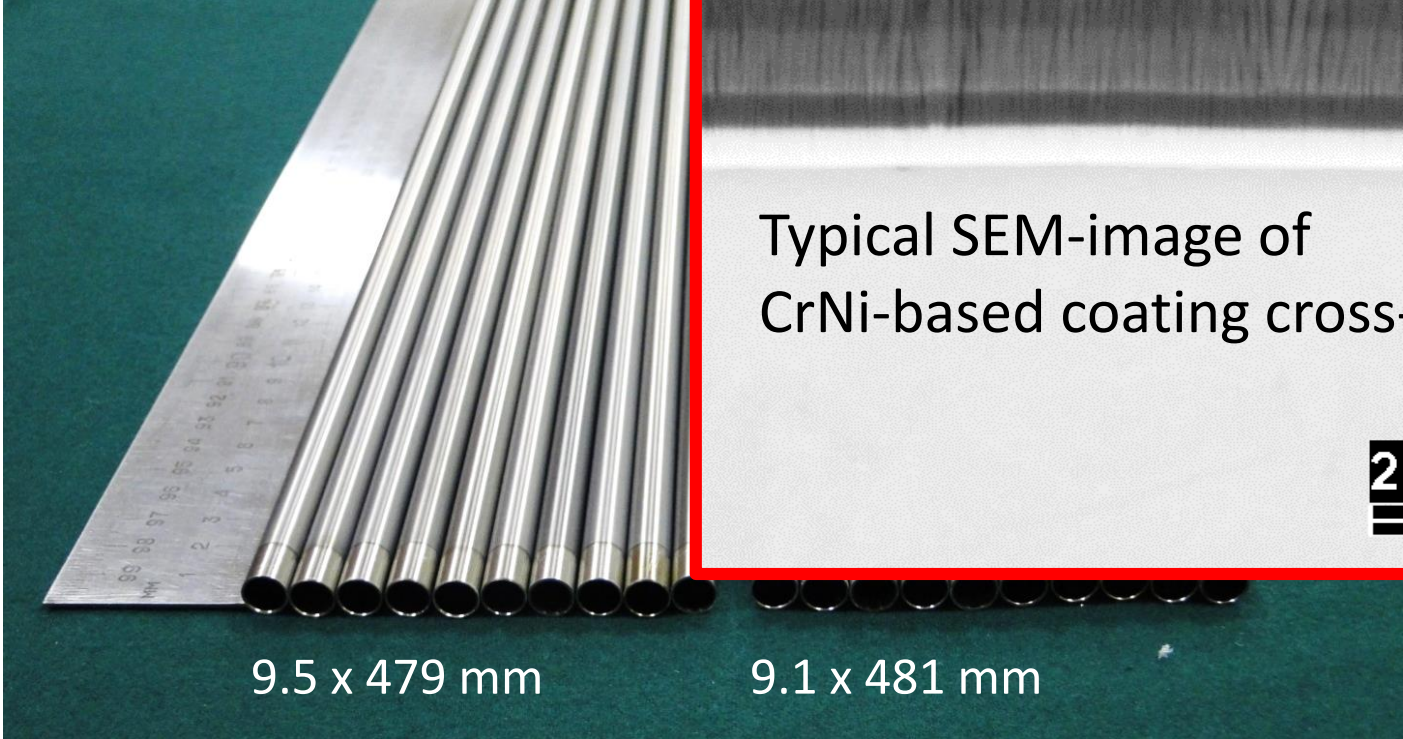
Coatings deposition



Coating base	Max thickness, μm
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NiCr	10 ± 1



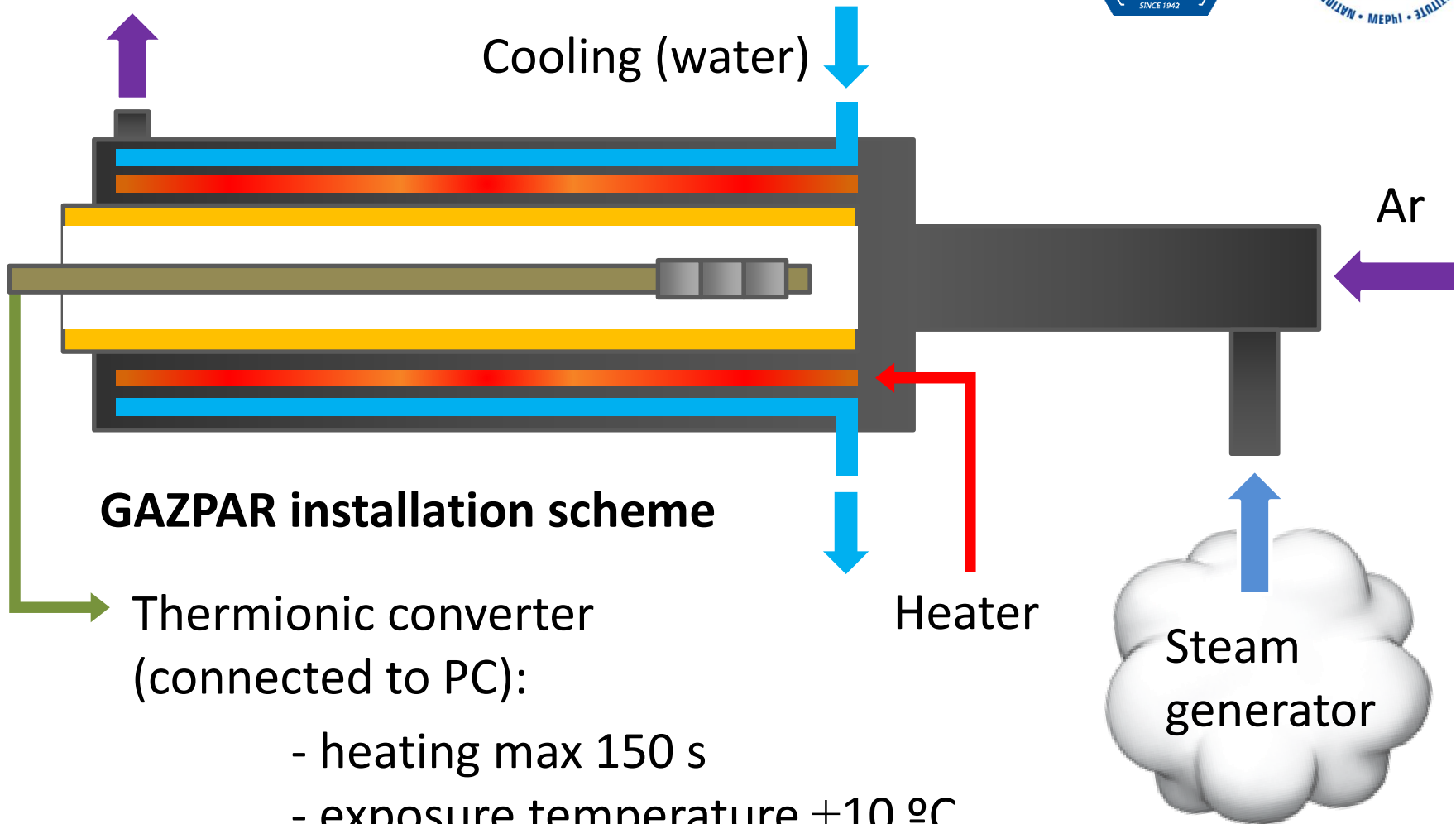
Typical SEM-image of CrNi-based coating cross-section



9.5 x 479 mm

9.1 x 481 mm

High-temperature corrosion

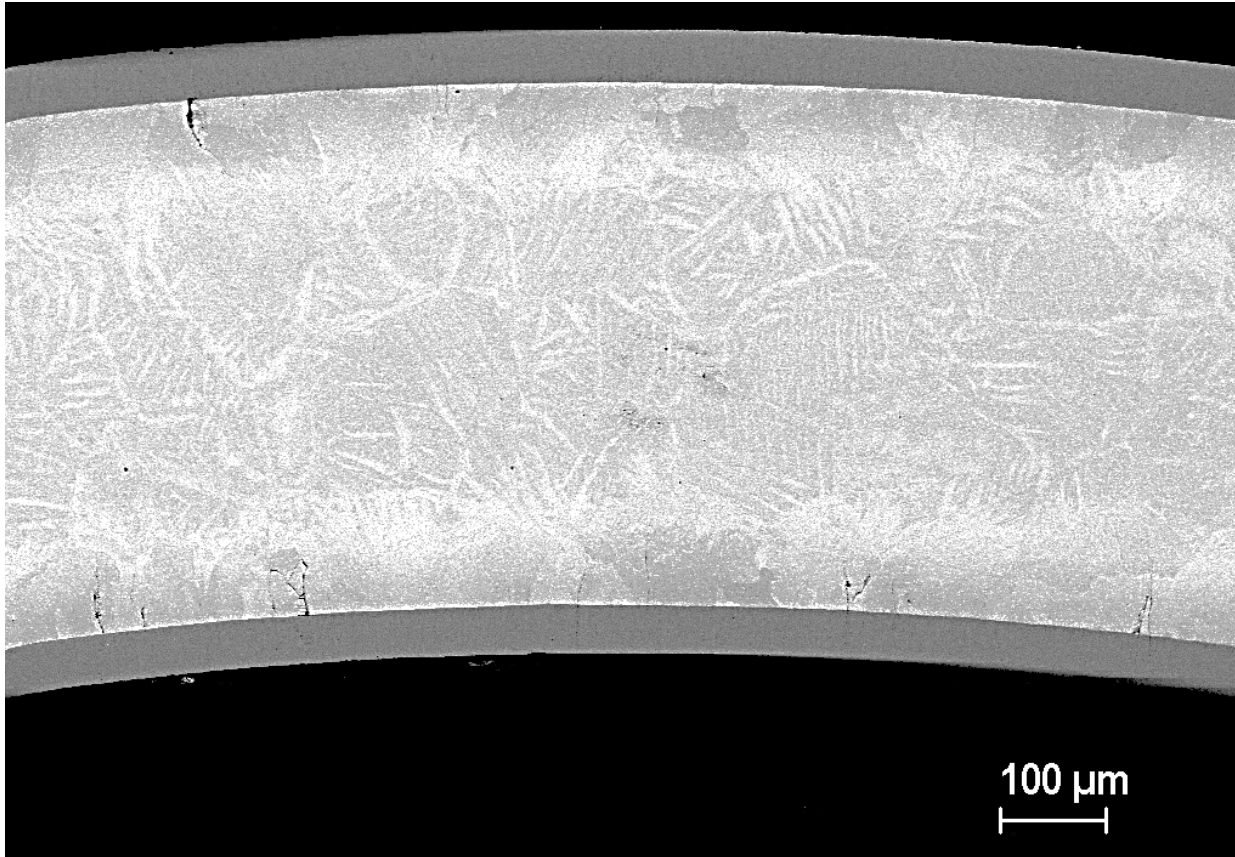


GAZPAR installation scheme

Thermionic converter
(connected to PC):

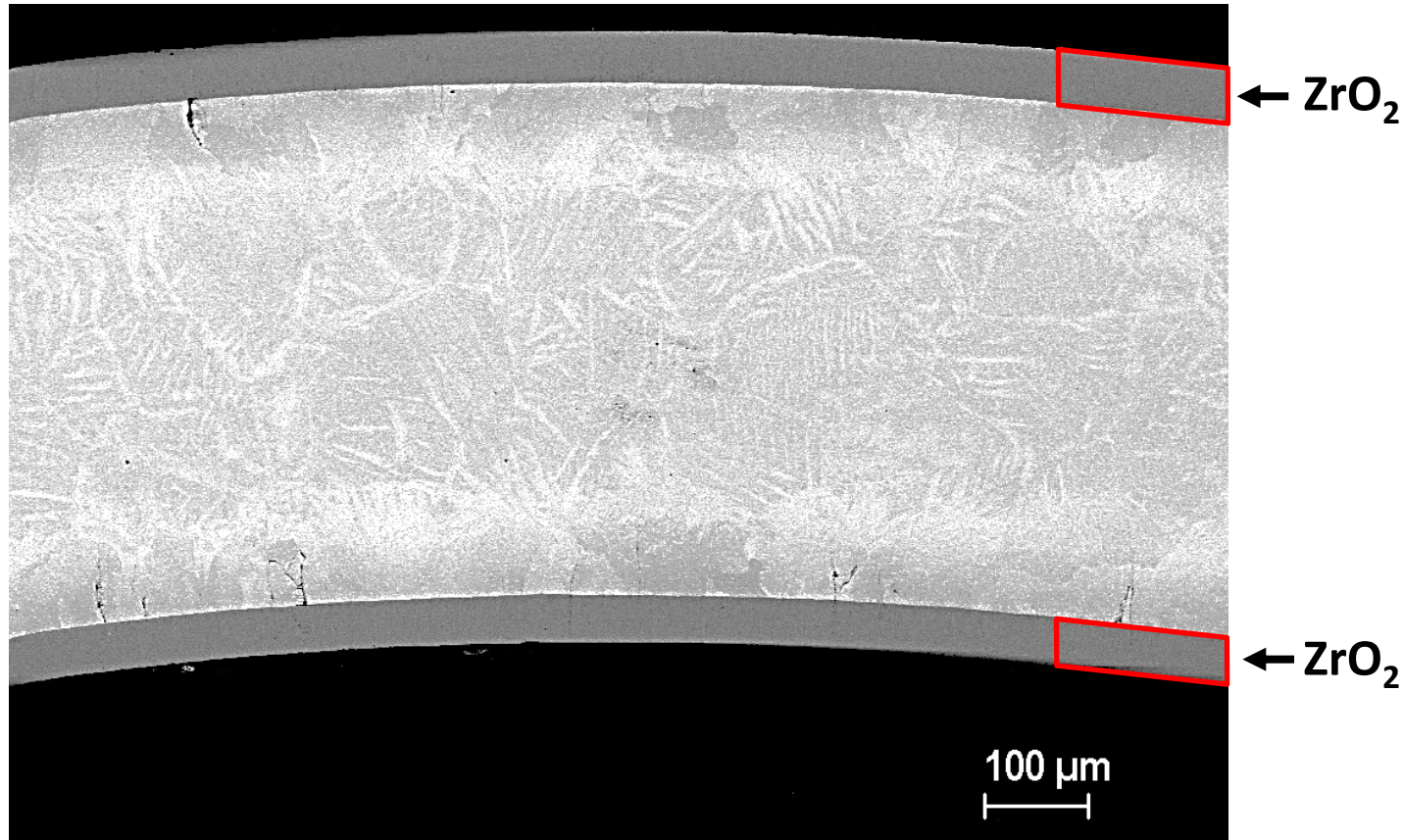
- heating max 150 s
- exposure temperature ± 10 °C
- cooling min 20 °C/s

Uncoated E110 corrosion



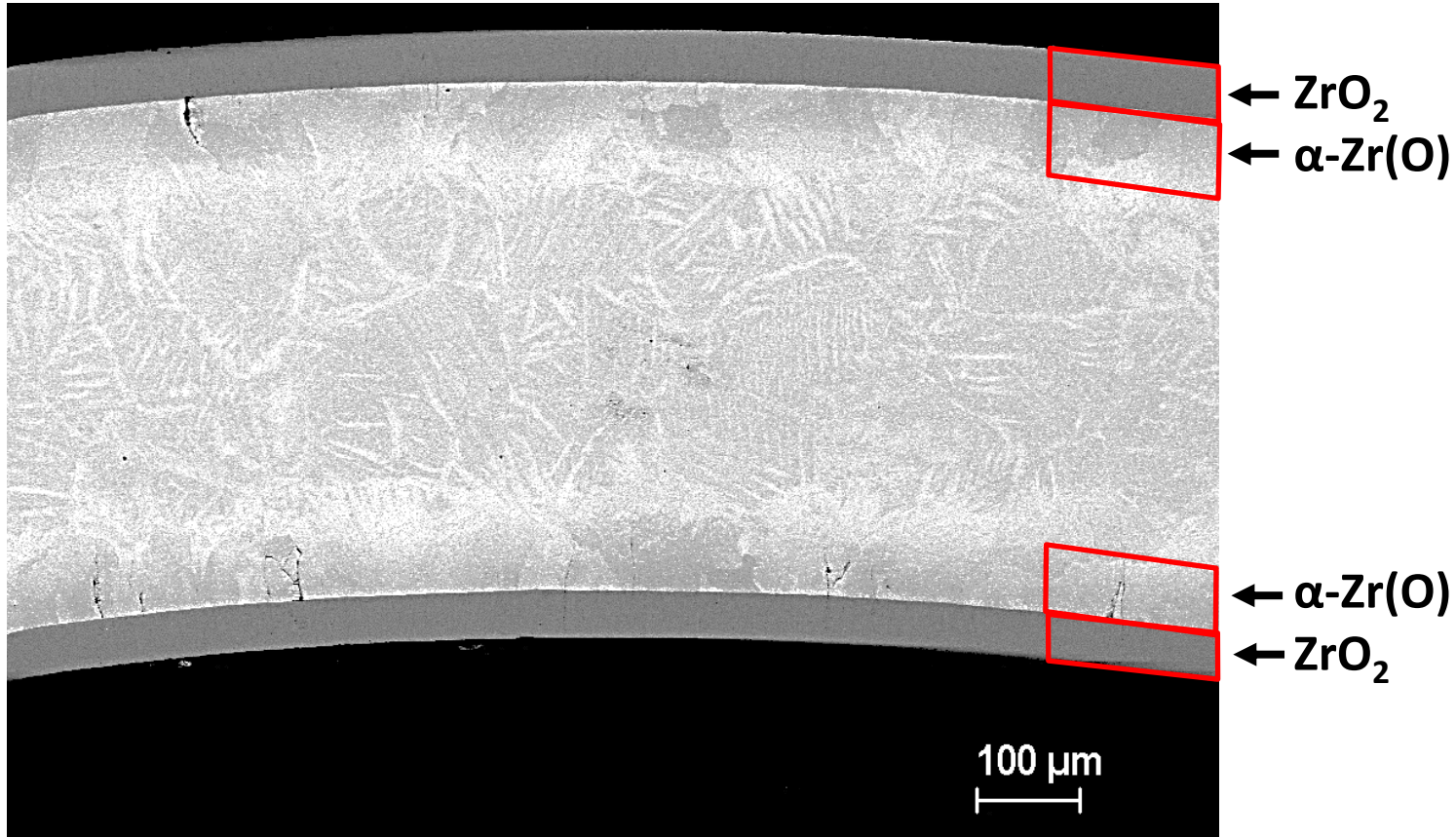
Typical SEM-image of E110 cladding cross-section after oxidation 1200 °C 500 s

Uncoated E110 corrosion



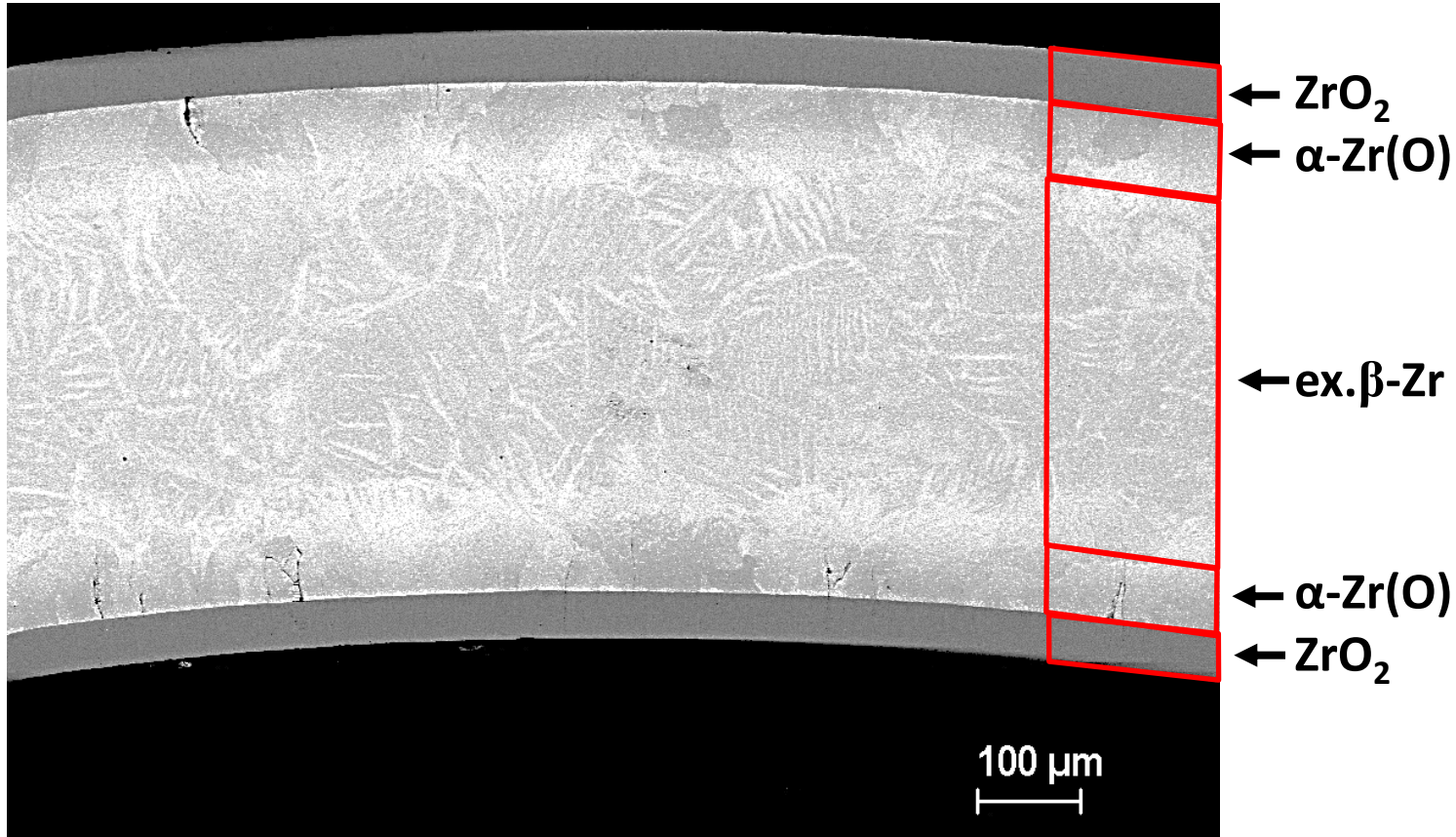
Typical SEM-image of E110 cladding cross-section after oxidation 1200 °C 500 s

Uncoated E110 corrosion



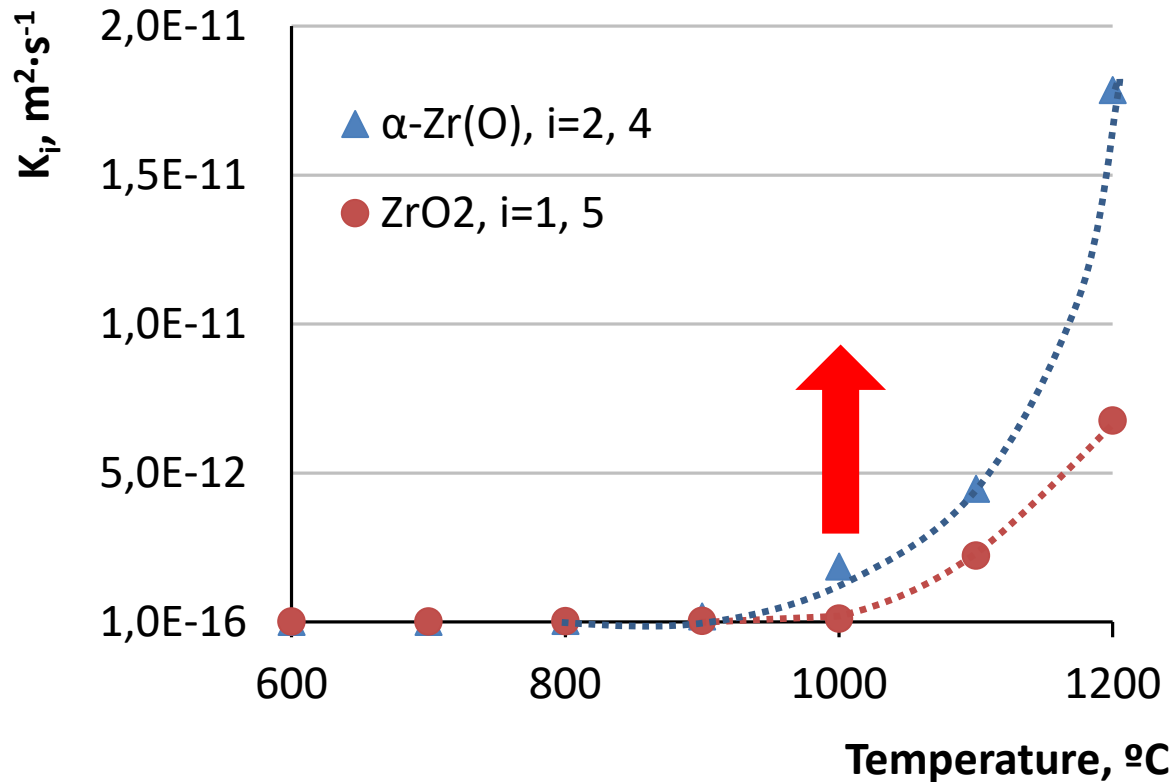
Typical SEM-image of E110 cladding cross-section after oxidation 1200 °C 500 s

Uncoated E110 corrosion



Typical SEM-image of E110 cladding cross-section after oxidation 1200 °C 500 s

Uncoated E110 corrosion



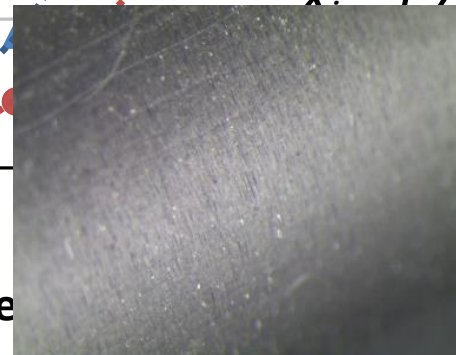
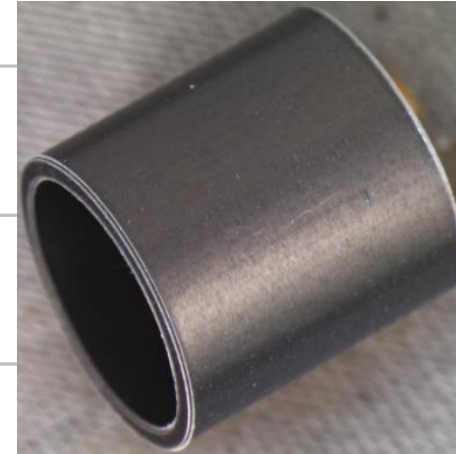
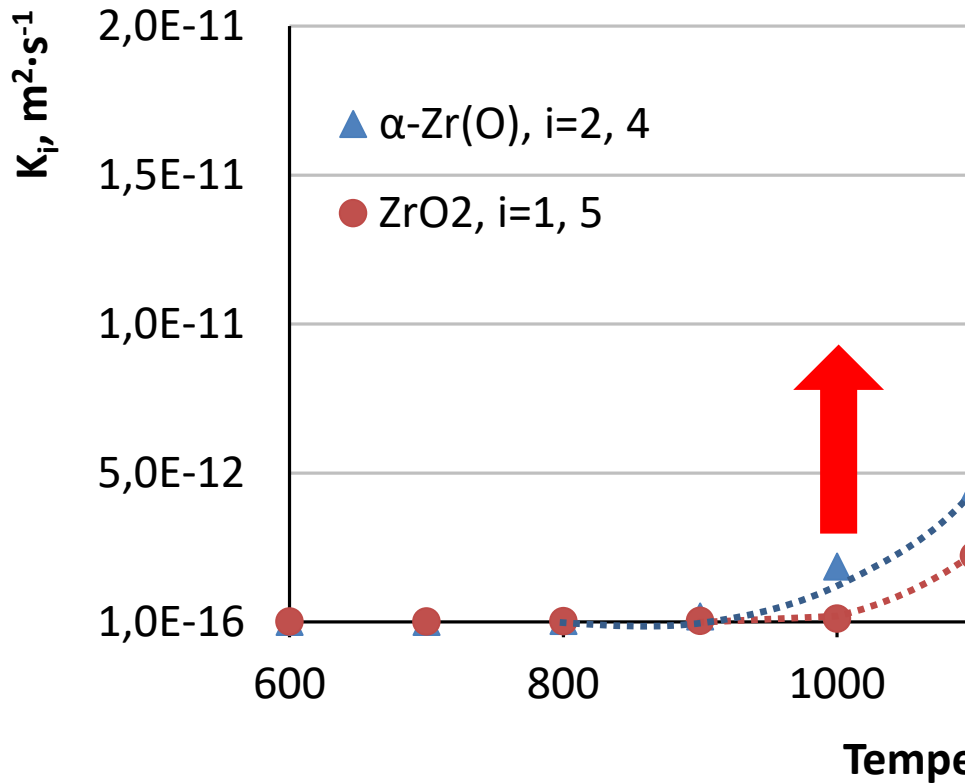
Phenomenological description:

$$x_i^2 = K_i \cdot t$$

x_i – i -zone thickness
 K_i – growth constant
 t – exposure time

Oxygen penetration rate increases about 10^4 - 10^5 times with increasing exposure temperature above 1000 $^{\circ}\text{C}$

Uncoated E110 corrosion



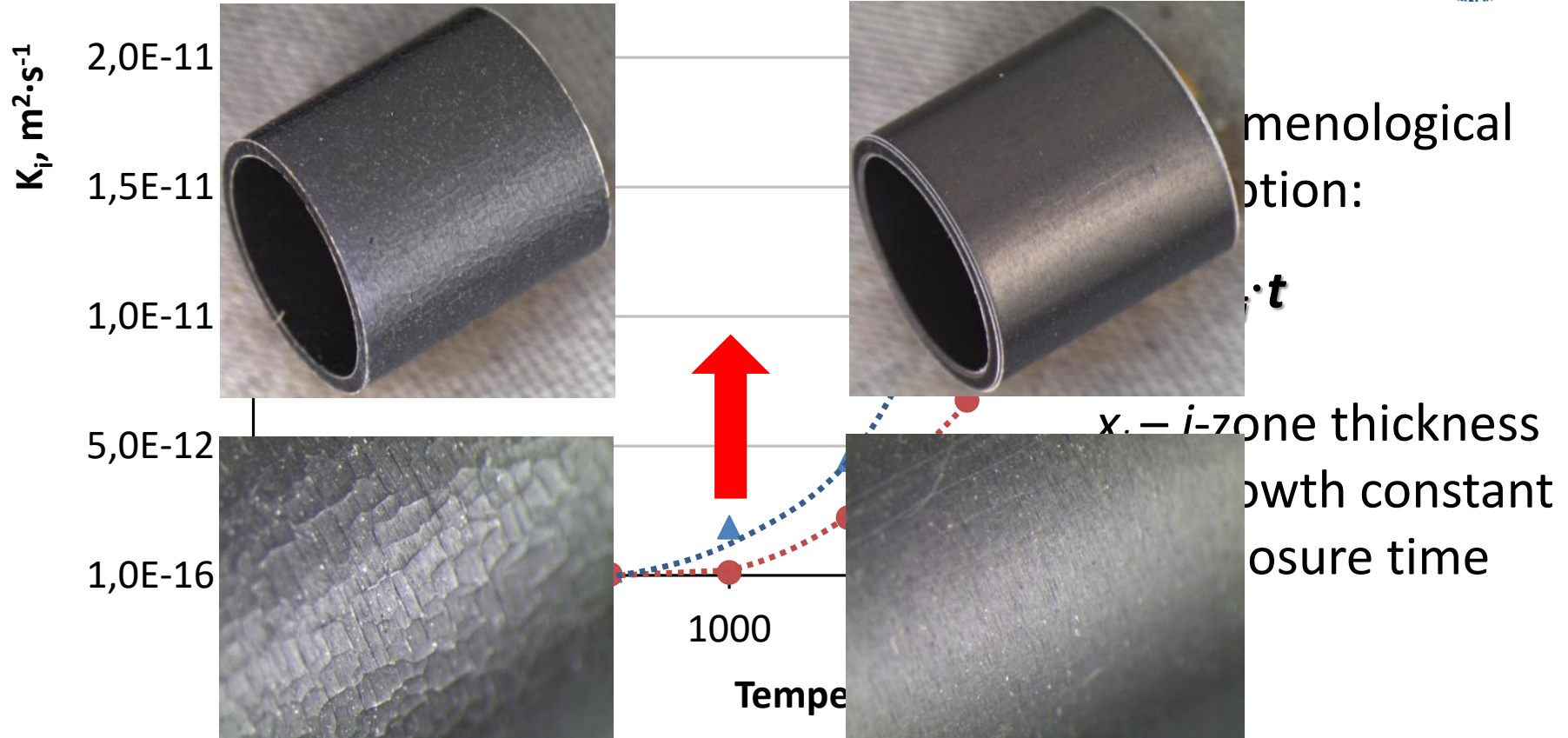
phenomenological
equation:

$x \propto t$

x – i -zone thickness
 k – growth constant
 t – exposure time

Uniform oxide
 1200 °C ECR = 27.9 %

Uncoated E110 corrosion



Cracks in oxide
800 °C ECR = 5.1 %

Uniform oxide
1200 °C ECR = 27.9 %

Cr-coatings at 800 °C



Oxidation conditions	Uncoated E110	Cr 2-4 μm	Cr 8-10 μm
800 °C 5000 s	ECR = 2.1 %	ECR = 1.5 %	ECR = 1.4 %
800 °C 7500 s	ECR = 3.3 %	ECR = 1.6 %	ECR = 2.3 %
1000 °C 5000 s	ECR = 5.3 %	ECR = 5.9 %	-

To study how **internal stresses caused by α - β transformation** effect on the chromium coatings preliminary experiments were carried out.

Cr-coatings at 800 °C



Oxidation conditions	Uncoated E110	Cr 2-4 μm	Cr 8-10 μm
800 °C 5000 s	ECR = 2.1 %	ECR = 1.5 %	ECR = 1.4 %
800 °C 7500 s	ECR = 3.3 %	ECR = 1.6 %	ECR = 2.3 %
1000 °C 5000 s	ECR = 5.3 %	ECR = 5.9 %	-



To study how **internal** effect on the chromium were carried out.

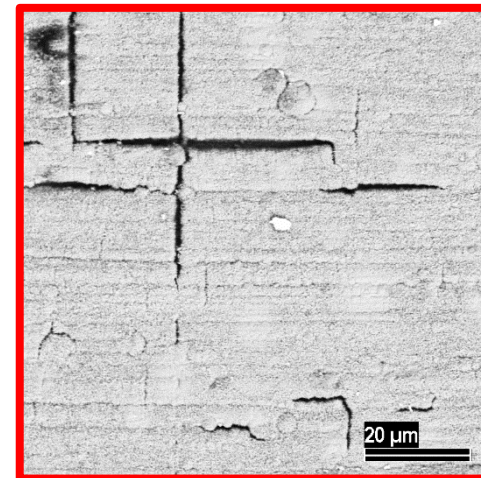
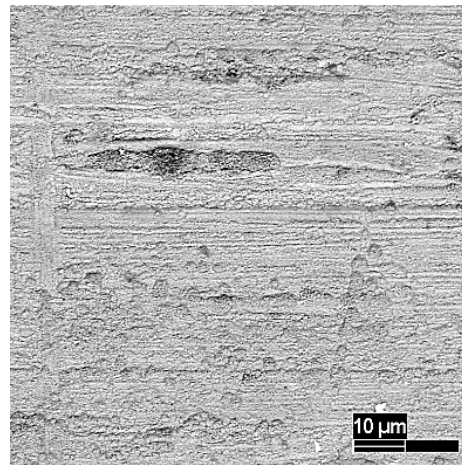


Cr-coatings at 800 °C



Oxidation conditions	Uncoated E110	Cr 2-4 μm	Cr 8-10 μm
800 °C 5000 s	ECR = 2.1 %	ECR = 1.5 %	ECR = 1.4 %
800 °C 7500 s	ECR = 3.3 %	ECR = 1.6 %	ECR = 2.3 %
1000 °C 5000 s	ECR = 5.3 %	ECR = 5.9 %	-

ECR increases due to **microcracks** formation

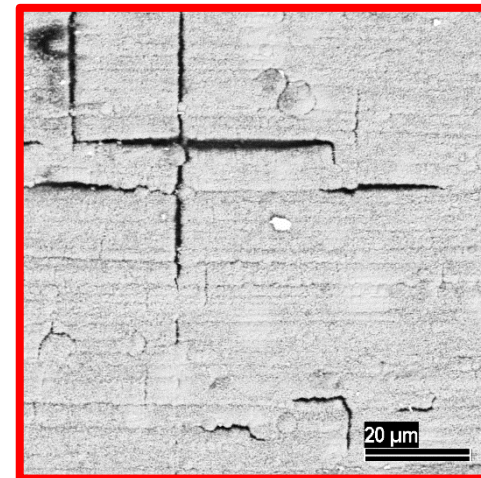
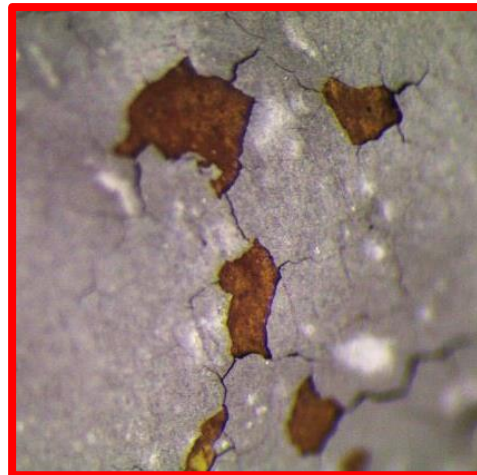


Cr-coatings at 800 °C



Oxidation conditions	Uncoated E110	Cr 2-4 μm	Cr 8-10 μm
800 °C 5000 s	ECR = 2.1 %	ECR = 1.5 %	ECR = 1.4 %
800 °C 7500 s	ECR = 3.3 %	ECR = 1.6 %	ECR = 2.3 %
1000 °C 5000 s	ECR = 5.3 %	ECR = 5.9 %	-

Cr-coating 2-4 μm is not enough to protect Zr



FeCrNi-coatings at 800 °C



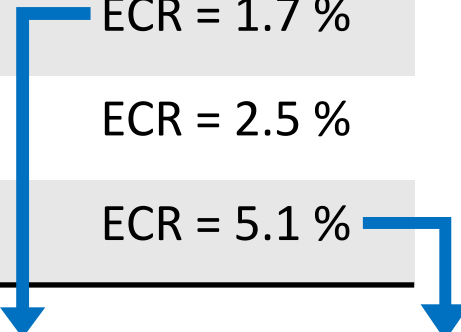
Oxidation conditions	Uncoated E110	FeCrNi 2-4 μm
800 °C 5000 s	ECR = 2.1 %	ECR = 1.7 %
800 °C 7500 s	ECR = 3.3 %	ECR = 2.5 %
1000 °C 5000 s	ECR = 5.3 %	ECR = 5.1 %

No destructions at the same conditions, but **lower efficiency** compare with Cr-coatings, especially at higher temperature.

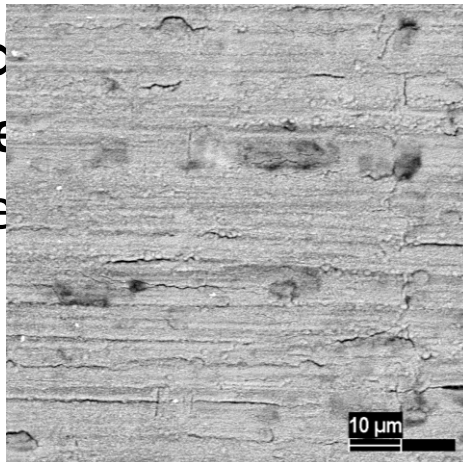
FeCrNi-coatings at 800 °C



Oxidation conditions	Uncoated E110	FeCrNi 2-4 μm
800 °C 5000 s	ECR = 2.1 %	ECR = 1.7 %
800 °C 7500 s	ECR = 3.3 %	ECR = 2.5 %
1000 °C 5000 s	ECR = 5.3 %	ECR = 5.1 %



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Cr/FeCrNi/Cr and Cr/CrNi/Cr

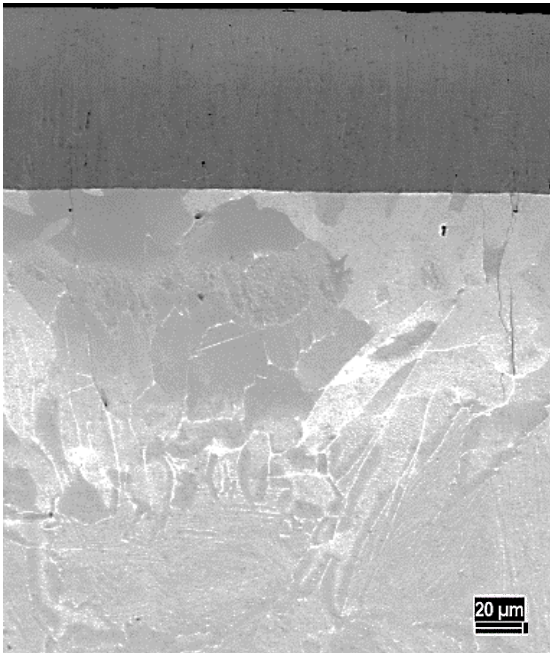


Composition	Thickness, μm	ECR, %
uncoated	-	17.3
Cr/FeCrNi/Cr	1.0 ± 0.3	17.4
Cr/FeCrNi/Cr	2.9 ± 0.3	18.4
Cr/FeCrNi/Cr	3.9 ± 0.3	19.0
Cr/FeCrNi/Cr	4.3 ± 0.3	19.0
Cr/FeCrNi/CrNi	7.0 ± 0.3	13.5
Cr/CrNi/Cr	1.0 ± 0.3	17.4
Cr/CrNi/Cr	4.5 ± 0.3	12.4
Cr/CrNi/Cr	6.5 ± 0.3	11.5
Cr/CrNi/Cr	8.7 ± 0.3	10.1
Cr/CrNi/Cr	9.8 ± 0.3	9.6
Cr/CrNi/Cr	10.8 ± 0.3	10.7

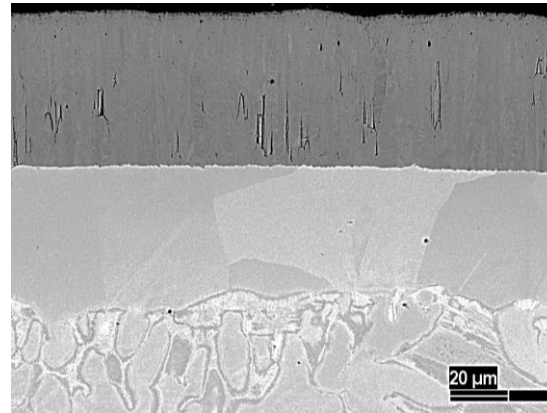
ECR calculated by weight gain shows obscure results

Oxidation at 1200 °C 400 s

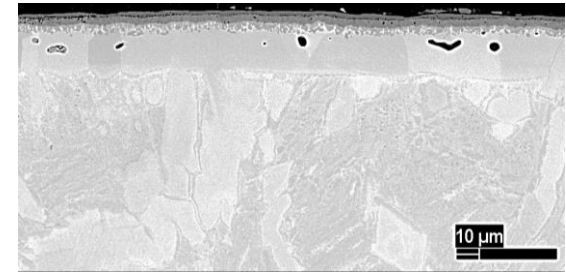
Cr/FeCrNi/Cr and Cr/CrNi/Cr



Uncoated E110
1200 °C 400 s

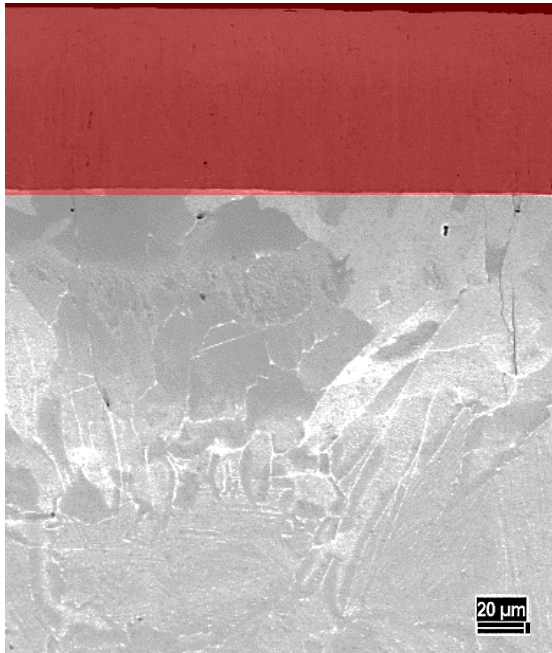


Cr/FeCrNi $4.1 \pm 0.3 \mu\text{m}$
1200 °C 400 s

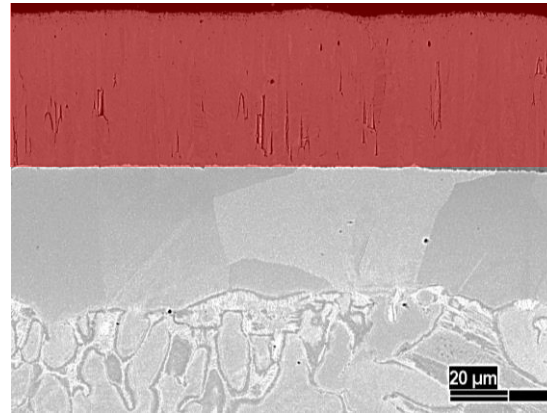


Cr/CrNi $9.8 \pm 0.3 \mu\text{m}$
1200 °C 400 s

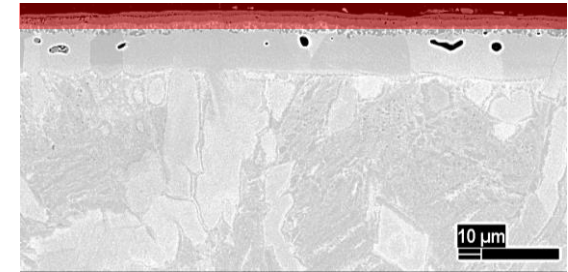
Cr/FeCrNi/Cr and Cr/CrNi/Cr



Uncoated E110
1200 °C 400 s
- ZrO₂ **50±5 μm**

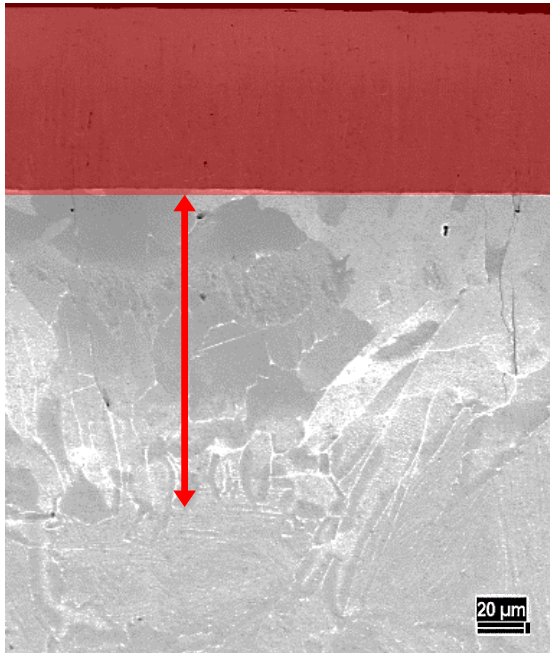


Cr/FeCrNi **4.1±0.3 μm**
1200 °C 400 s
- slightly affect on
oxide thickness **10%**

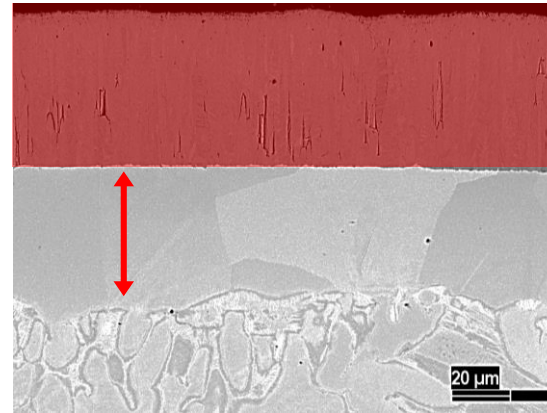


Cr/CrNi **9.8±0.3 μm**
1200 °C 400 s
- decrease oxide
zone significantly **70%**

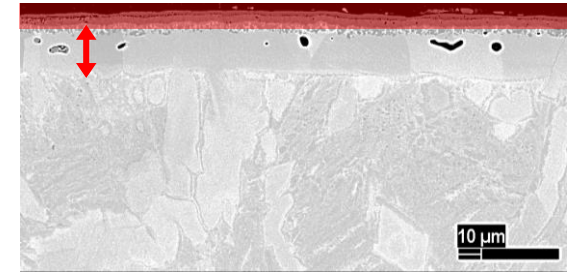
Cr/FeCrNi/Cr and Cr/CrNi/Cr



Uncoated E110
1200 °C 400 s
- ZrO₂ 50±5 μm
- α-Zr(O) **80±10 μm**

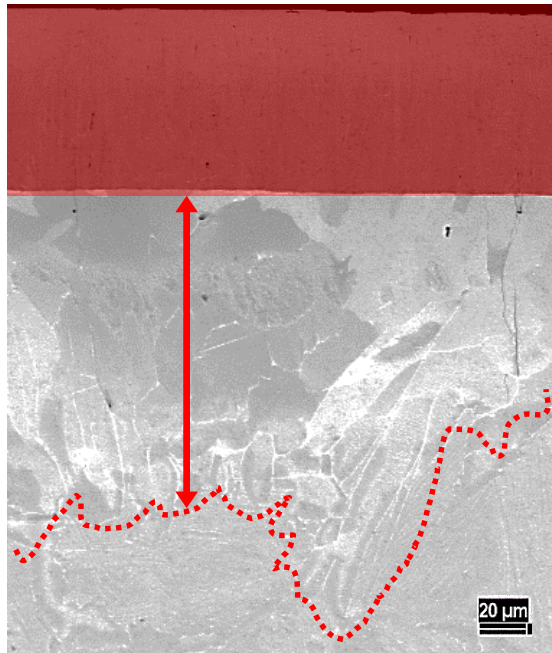


Cr/FeCrNi 4.1±0.3 μm
1200 °C 400 s
- slightly affect on
oxide thickness **10%**
- decrease α-Zr(O)
layer size **30%**

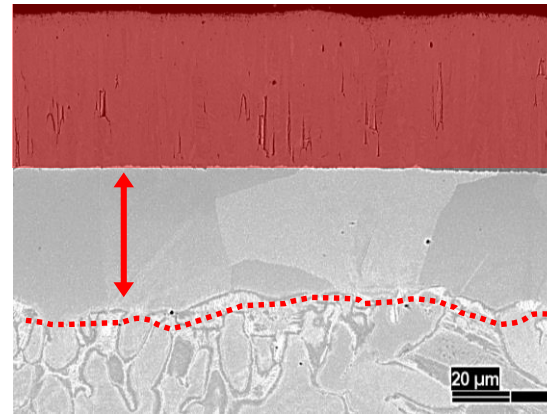


Cr/CrNi 9.8±0.3 μm
1200 °C 400 s
- decrease oxide
zone significantly **70%**
- decrease α-Zr(O)
layer size **60%**

Cr/FeCrNi/Cr and Cr/CrNi/Cr

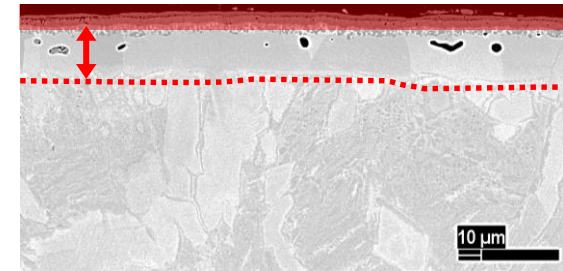


Uncoated E110
 1200 °C 400 s
 - ZrO₂ 50±5 μm
 - α-Zr(O) 80±10 μm



Cr/FeCrNi 4.1±0.3 μm
 1200 °C 400 s

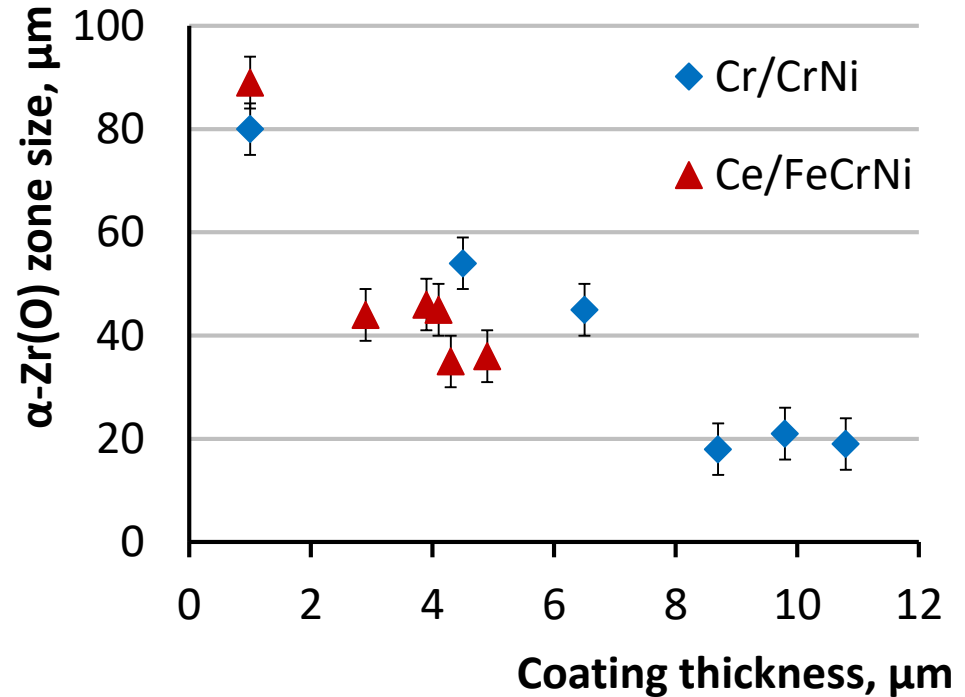
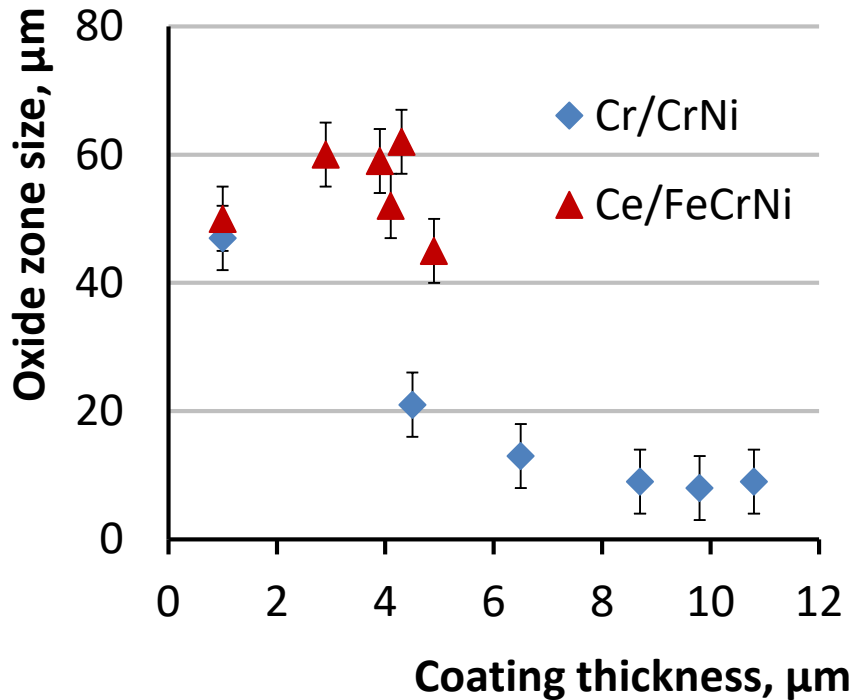
- slightly affect on oxide thickness **10%**
- decrease α-Zr(O) layer size **30%**
- make **α-Zr(O) | ex.β-Zr** interface smooth



Cr/CrNi 9.8±0.3 μm
 1200 °C 400 s

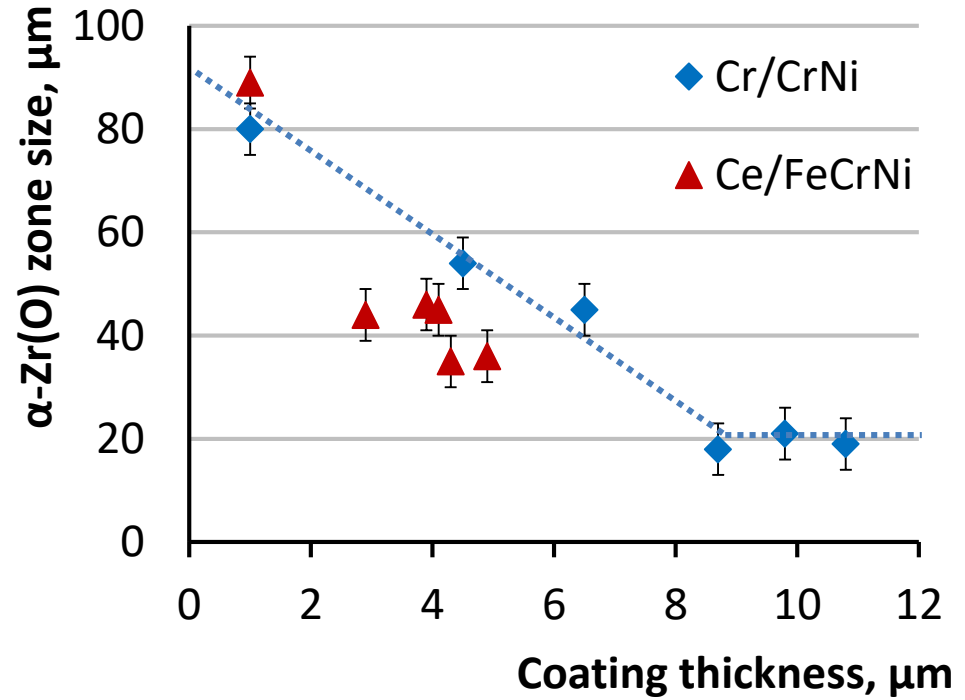
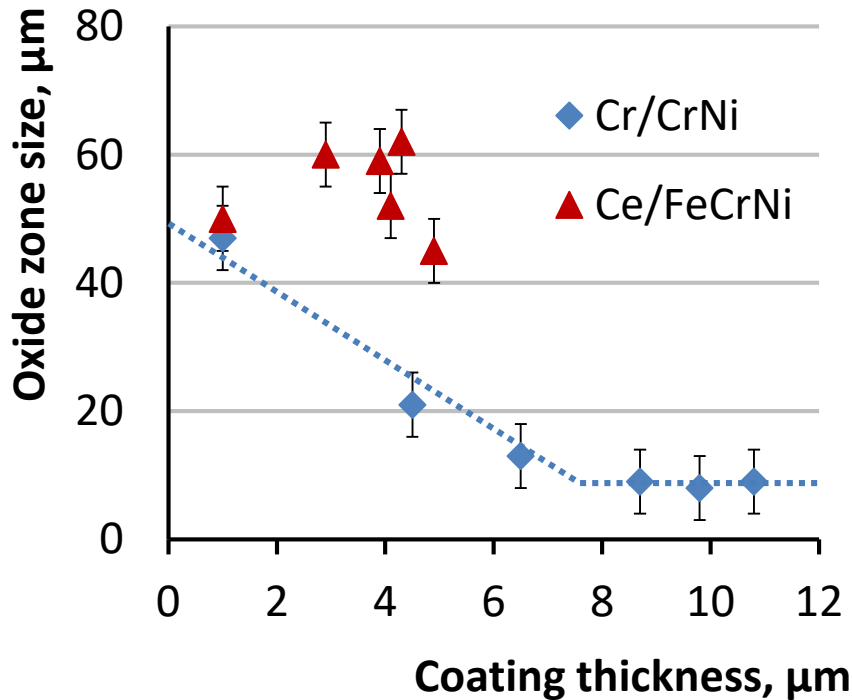
- decrease oxide zone significantly **70%**
- decrease α-Zr(O) layer size **60%**
- make **α-Zr(O) | ex.β-Zr** interface smooth

Coating thickness effect at 1200 °C 400 s



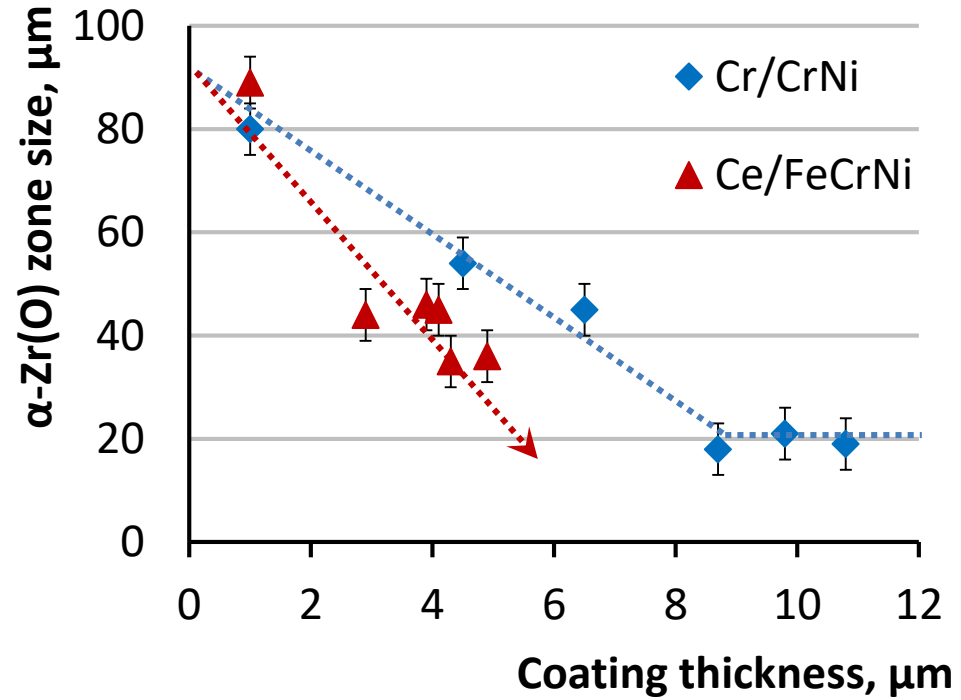
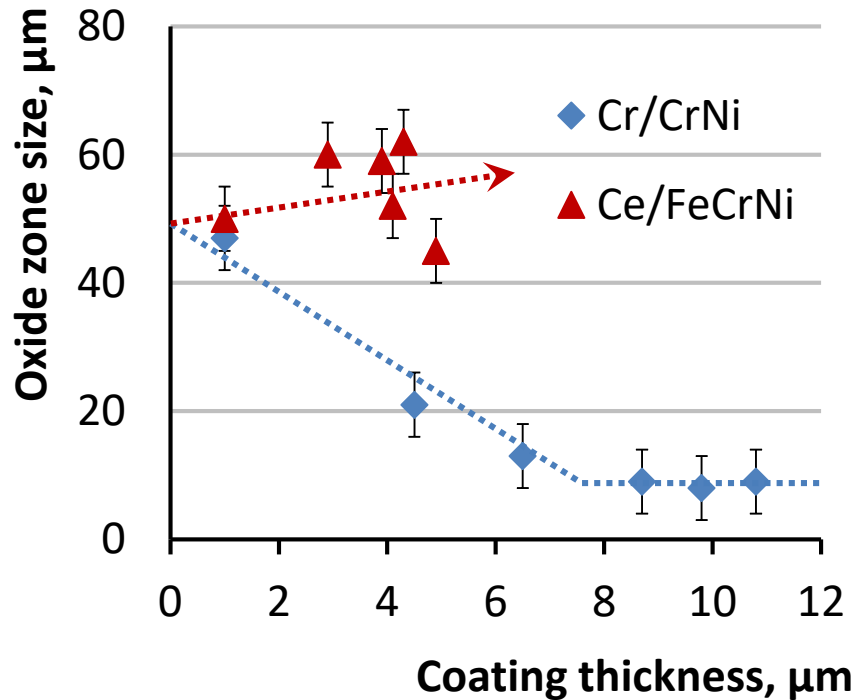
Value «oxide zone» on this figure include ZrO_2 layer and oxides formed at coatings deposited.

Coating thickness effect at 1200 °C 400 s



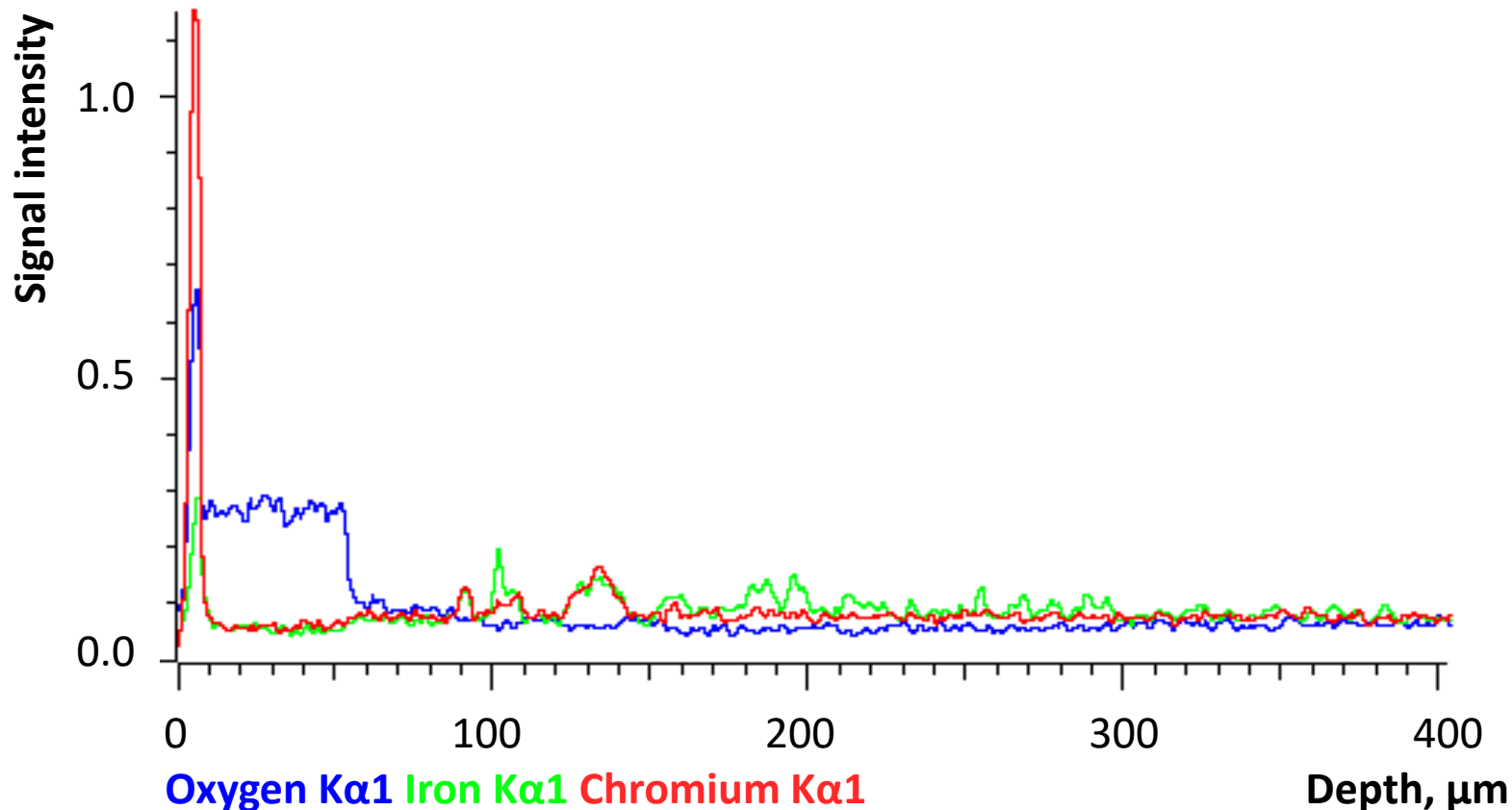
Oxygen penetration degree decreases with increasing Cr/CrNi coating thickness and saturates at about 8 μm (blue dotted lines).

Coating thickness effect at 1200 °C 400 s



Samples with Cr/FeCrNi coatings reveal higher oxide zone size, but lower size of α -Zr(O) zone (red dotted arrows).

X-ray microanalysis

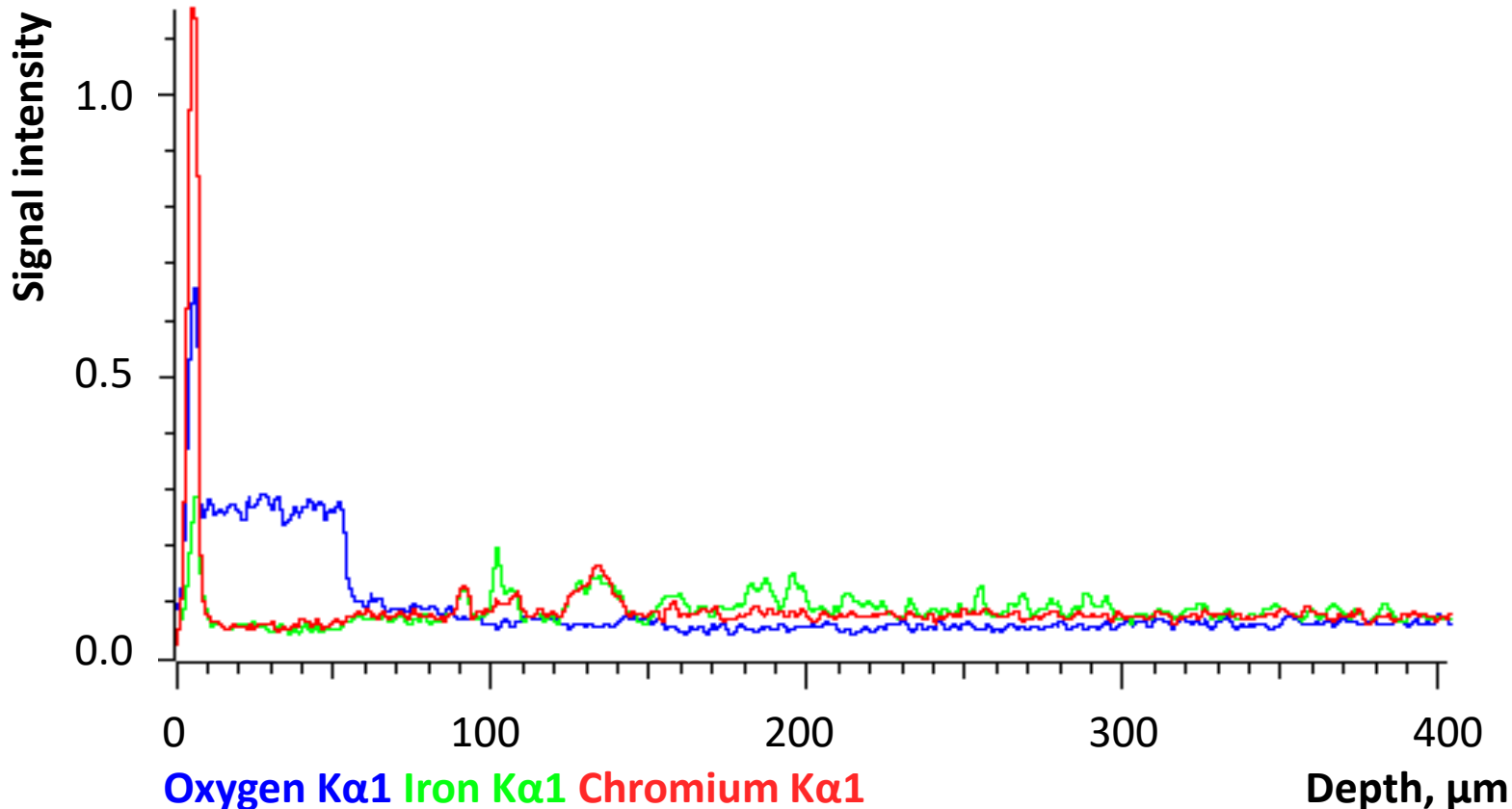


Data from sample with Cr/FeCrNi coating 4.1 μm thick after oxidation at 1200 $^{\circ}\text{C}$ 400 s

X-ray microanalysis



↓ Cr/FeCrNi coating oxidized

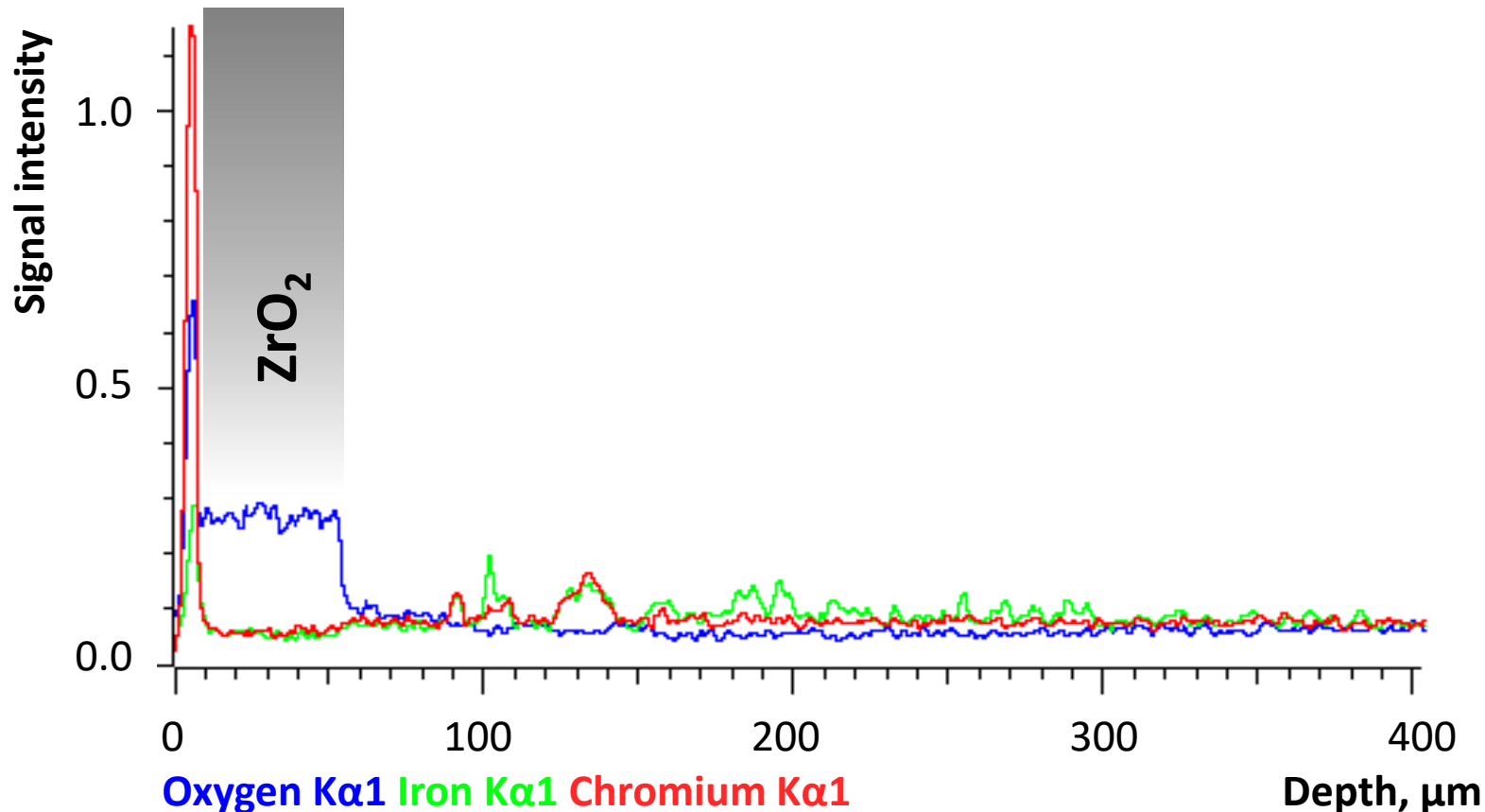


Data from sample with Cr/FeCrNi coating 4.1 μm thick after oxidation at 1200 °C 400 s

X-ray microanalysis



↓ Cr/FeCrNi coating oxidized

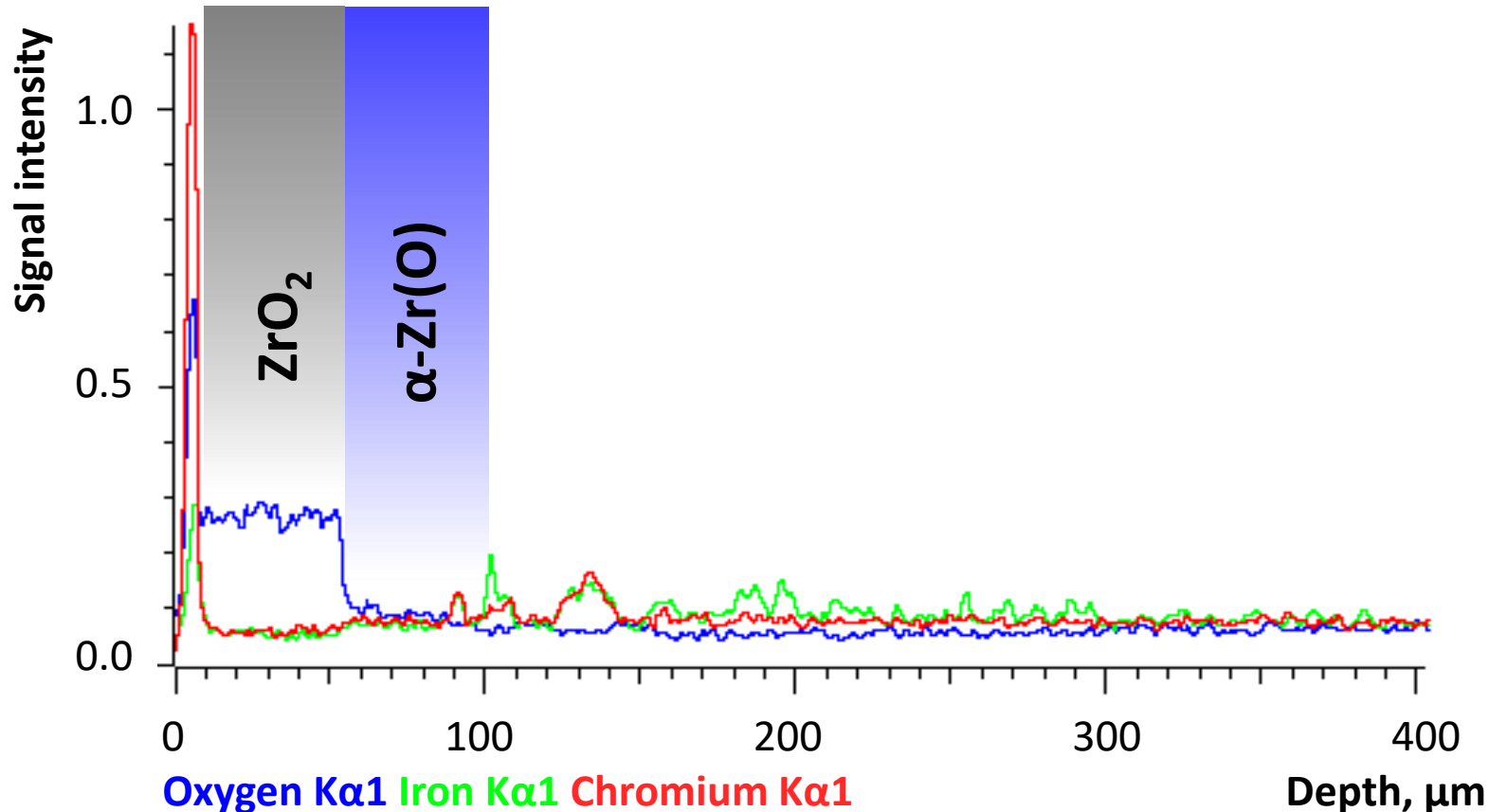


Data from sample with Cr/FeCrNi coating 4.1 μm thick after oxidation at 1200 °C 400 s

X-ray microanalysis



↓ Cr/FeCrNi coating oxidized

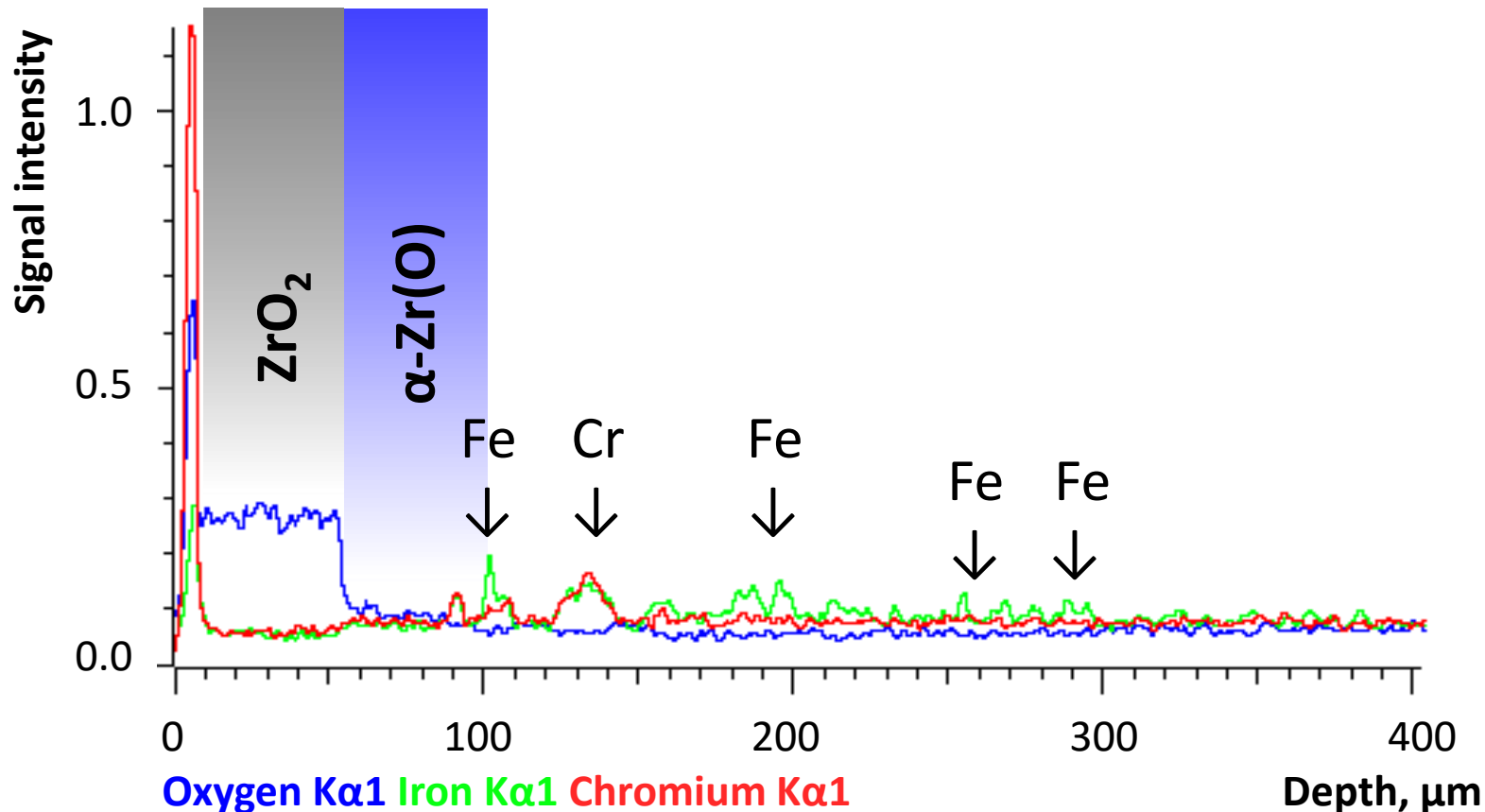


Data from sample with Cr/FeCrNi coating 4.1 μm thick after oxidation at 1200 °C 400 s

X-ray microanalysis



↓ Cr/FeCrNi coating oxidized

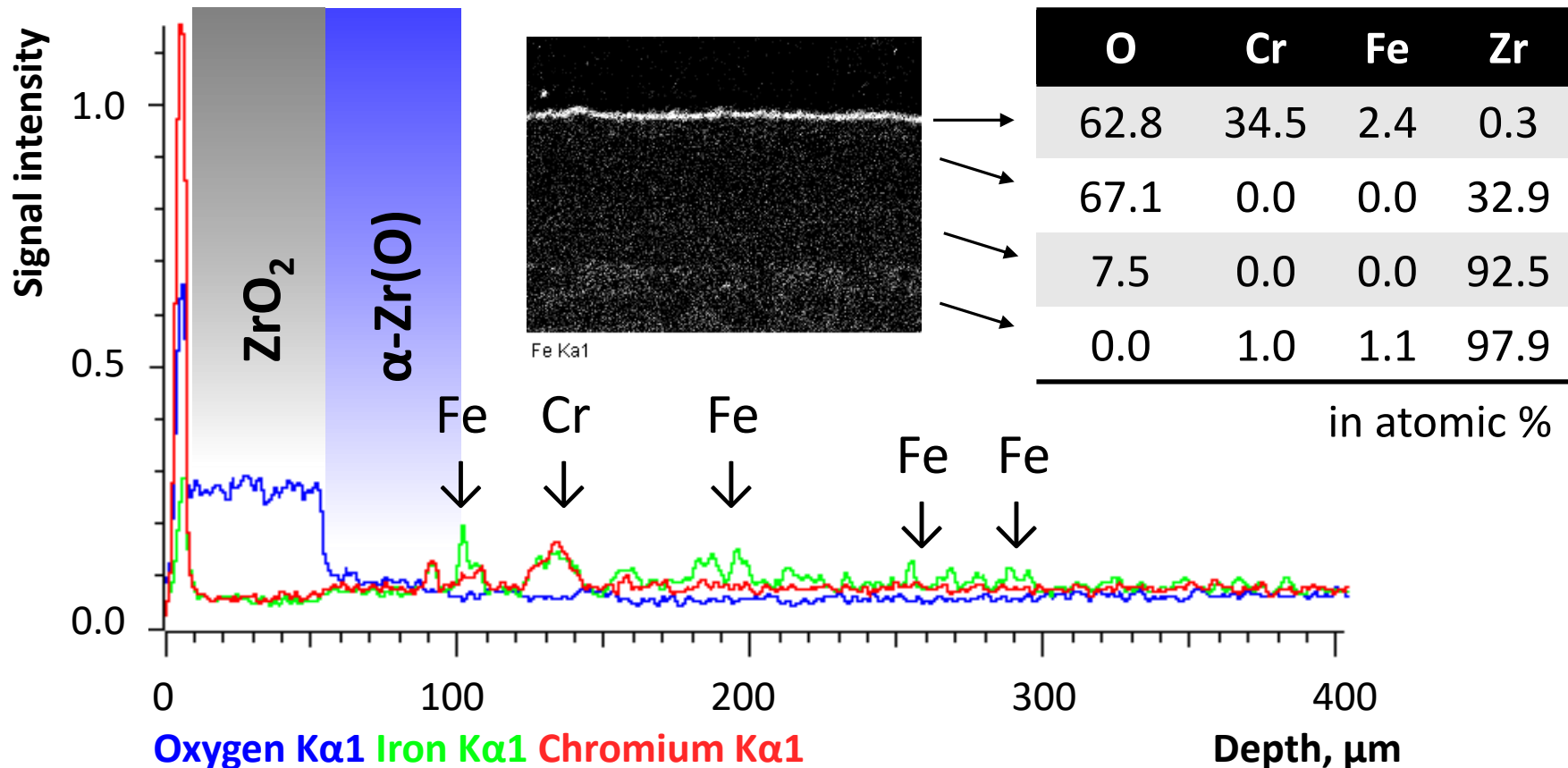


Data from sample with Cr/FeCrNi coating 4.1 μm thick after oxidation at 1200 °C 400 s

X-ray microanalysis

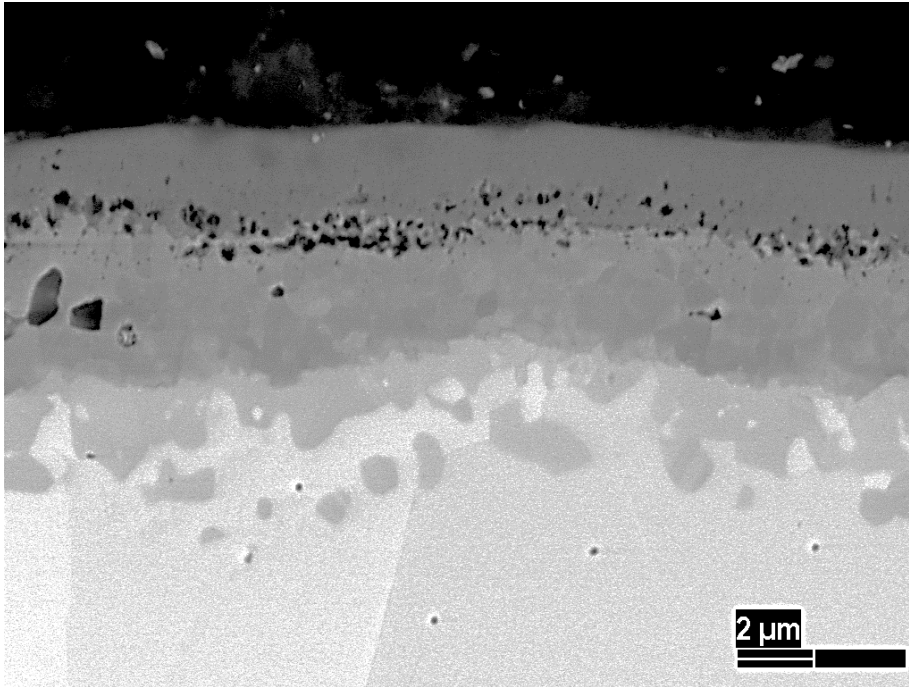


↓ Cr/FeCrNi coating oxidized



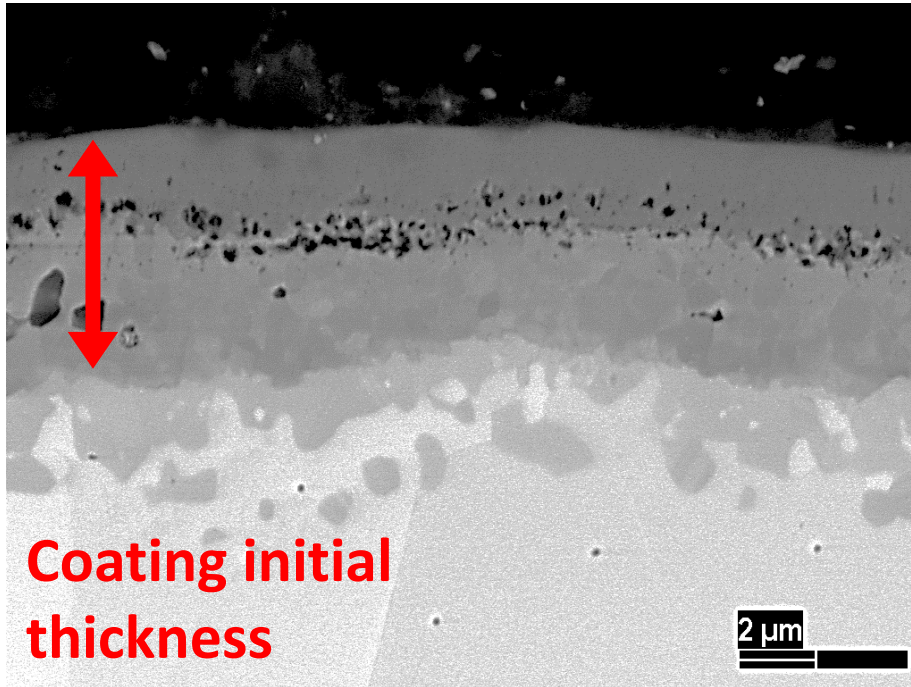
Data from sample with Cr/FeCrNi coating 4.1 μm thick after oxidation at 1200 °C 400 s

Cr/CrNi coating cross section



Multilayer structure of oxidized Cr/CrNi coating is observed (oxidation at 1200 °C 400 s)

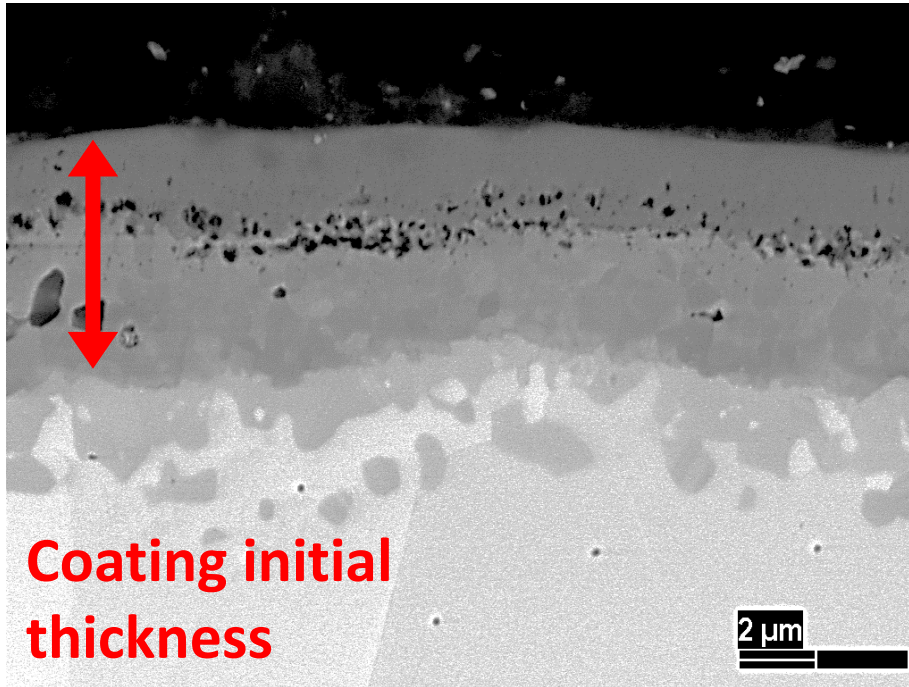
Cr/CrNi coating cross section



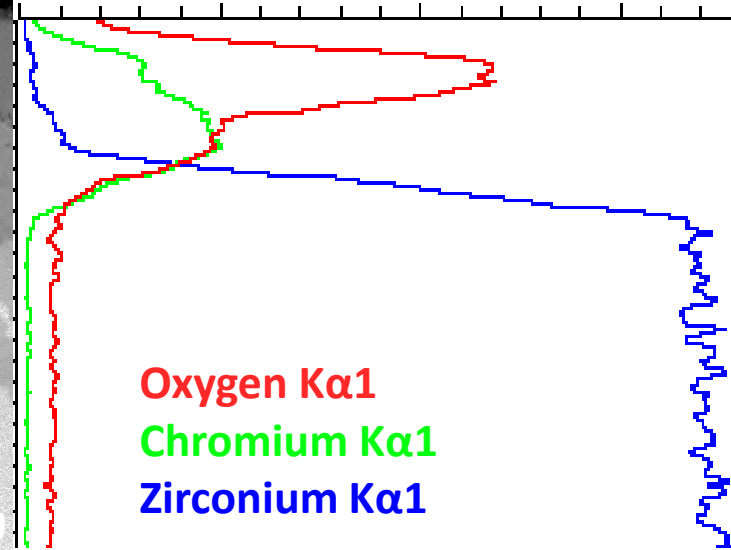
**Coating initial
thickness**

Multilayer structure of oxidized Cr/CrNi coating is observed (oxidation at 1200 °C 400 s)

Cr/CrNi coating cross section

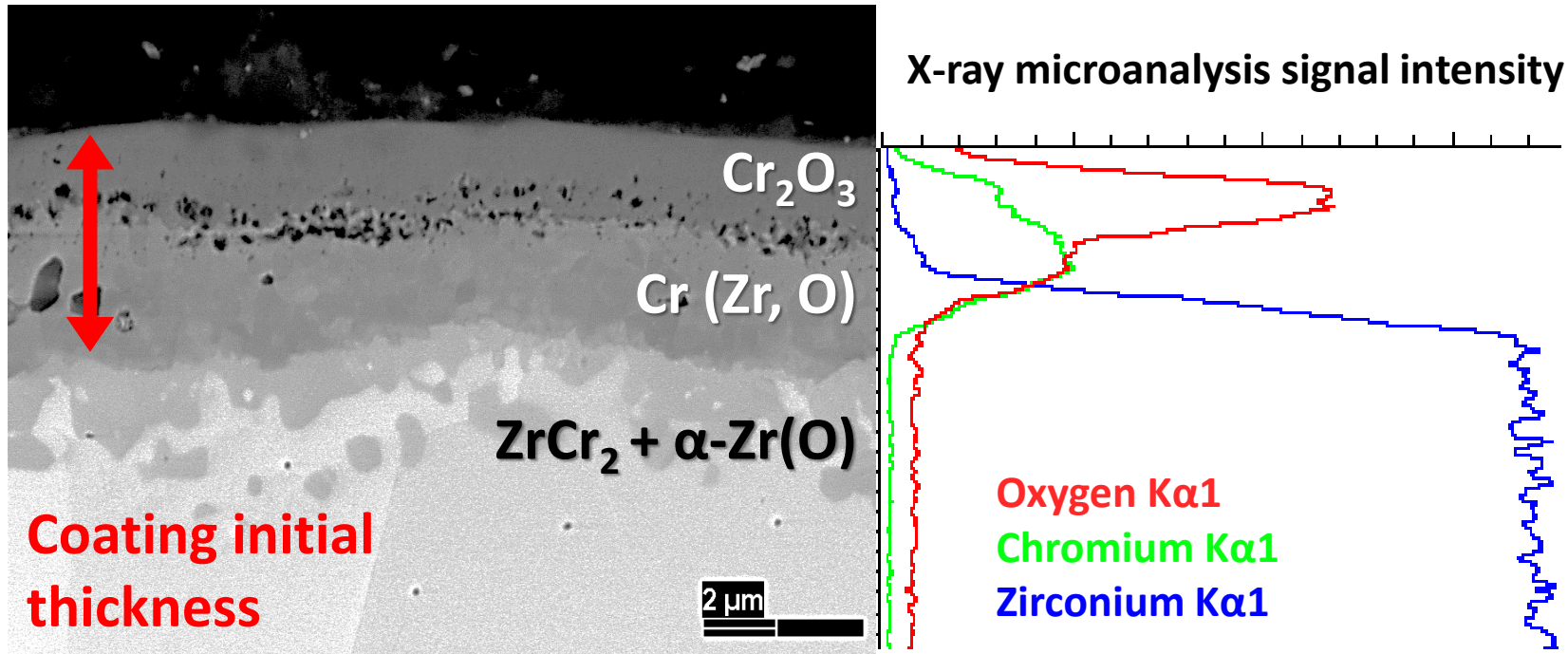


X-ray microanalysis signal intensity



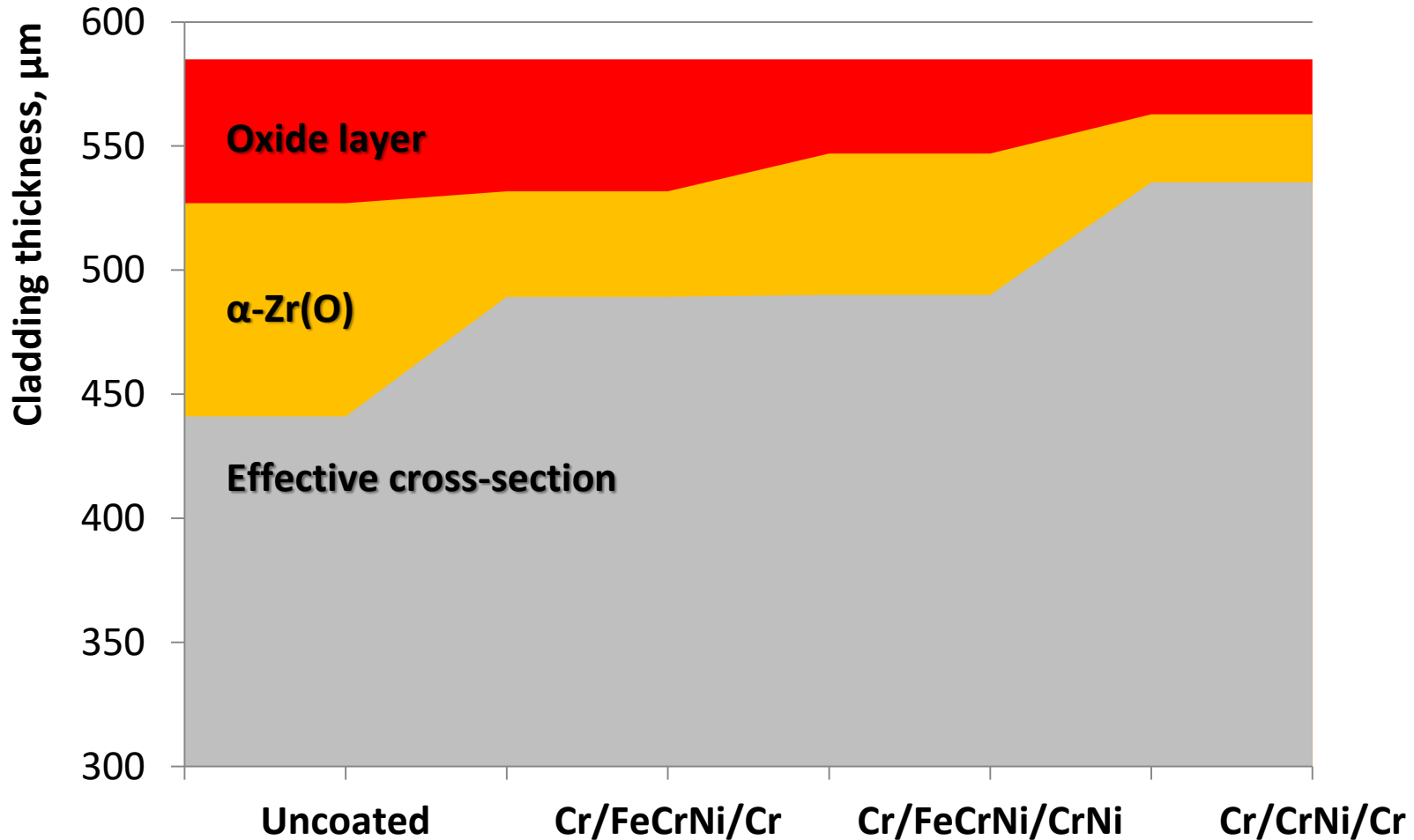
Multilayer structure of oxidized Cr/CrNi coating is observed (oxidation at 1200 °C 400 s)

Cr/CrNi coating cross section



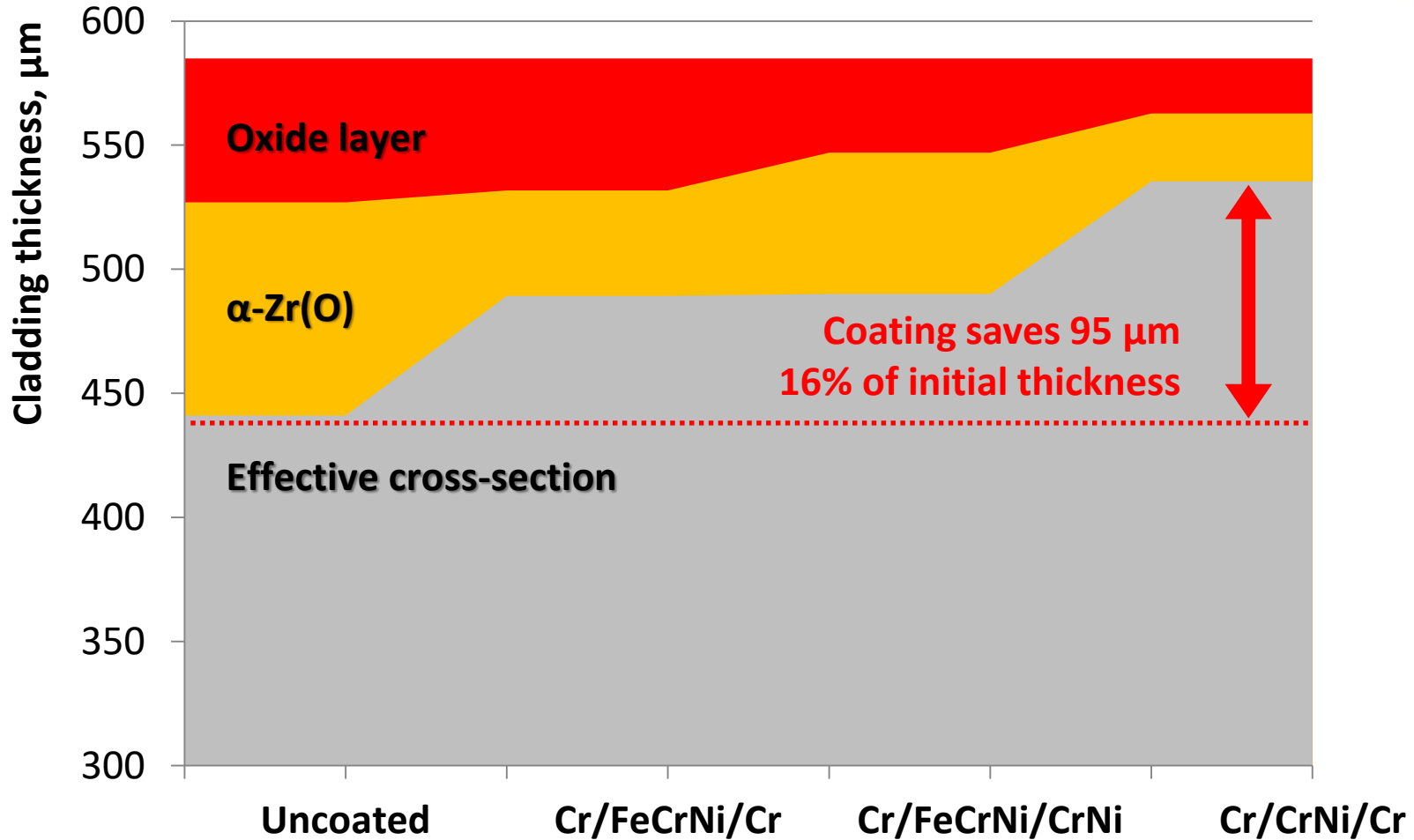
It is observed that **diffusion greatly affect** coatings evolution during oxidation at $1200\ \text{°C}$

Effective cladding cross section



Oxidation at 1200 °C 400 s

Effective cladding cross section



Oxidation at 1200 °C 400 s

Conclusion



1. Experimental results on the high-temperature oxidation of fuel claddings fragments from E110 alloy with FeCrNi and CrNi based multilayer coatings 1.0-10.0 μm thick were obtain;
2. It is shown that **microcracks** arises in Cr-coatings during oxidation at 800 $^{\circ}\text{C}$ presumably caused by internal stresses α - β transformation in Zr may lead to coatings destruction and increase corrosion;
3. It is established that Fe and Cr atoms migrate from coatings to α -Zr(O) | ex. β -Zr interface during high-temperature oxidation, that presumably control oxygen penetration into the bulk metal;
4. It is shown that diffusion fast rate leads to coatings stratification during high-temperature oxidation;
5. As observed, Cr/FeCrNi/Cr and Cr/CrNi/Cr coatings deposited reduce total oxygen penetration depth by **34%** and **66%** respectively during oxidation at 1200 $^{\circ}\text{C}$, save up to **95 μm** of cladding material.



Thank you for attention!